



International Journal of PharmTech Research CODEN (USA): IJPRIF ISSN : 0974-4304 Vol. 3, No.1, pp 153-156, Jan-Mar 2011

Spectrophotometric Method Development and Determination of Ornidazole in Bulk and Tablet Dosage Form

Rana Mazumder*¹, Lila K. Nath², Tapan K. Giri³, Prasanta K. Choudhury⁴,

Ayan K. Kar,¹ and Manas K. Sarkar¹

¹Department of Pharmaceutical Technology, Calcutta Institute of Pharmaceutical Technology and Allied Health Sciences, Howrah-711316, West Bengal, India.

²Professor, Department of Pharmaceutical Sciences, Dibrugarh University, Dibrugarh, Assam-786004, India.

³Associate Professor, Rungta College of Pharmaceutical Sciences and Research, Vhilai-491024, India.

⁴Department of Pharmaceutical Technology, Royal College of Pharmacy, Berhampur, Ganjam, Orissa-760002, India.

*Corres.author: rm_dreamsin@indiatimes.com, Mobile No.09434374834

Abstract: Ornidazole is an antimicrobial agent. Here, to develop a simple, precise and accurate spectrophotometric method determination and validation for ornidazole in bulk and solid dosage form. Ornidazole shows a maximum absorbance at 310.5 nm in ethanol as a medium. Beer's law was observed in the concentration range of 5-25 μ g/ml. The methods were validated statistically and by recovery studies and it were found to be accurate, precise and reproducible for determination of ornidazole in bulk and marketed dosage form. The methods were validated statistically and by recovery studies and reproducible for determination of ornidazole in bulk and marketed dosage form. The methods were validated statistically and by recovery studies and reproducible for determination of ornidazole in bulk and solid dosage form.

Key words: Ornidazole, Ethanol, Spectrophotometric Method, Validation.

INTRODUCTION

Ornidazole is a 5-nitroimidazole derivative, chemically-chloromethyl-2-methyl-5nitro-1H-

imidazole-1-ethanol with molecular formula C7H1 0N3O3Cl¹ used as antimicrobial agent. Ornidazole has antiprotozoal and antibacterial properties against anaerobic bacteria. The antimicrobial activity of this compound is due to reduction of the nitro group to a more reactive amine group that attacks microbial DNA, inhibiting further synthesis, and leading to degradation of existing DNA.² It is not official in any pharmacopoeia. Literature survey³⁻⁸ reveals that ornidazole is estimated by voltametry 9 and HPLC10 methods for its determination in dosage forms and biological fluids.

Ornidazole, chemically α -chloromethyl-2-methyl-5nitro-1H-imidazole-1-ethanol, with molecular formula $C_7H_{10}N_3O_3Cl$. It is not an official drug in any pharmacopoeia. Few methods are available for the determination of ornidazole in bulk and solid dosage form. The present studies describe accurate, precise and reproducible spectrophotometric methods for estimation of ornidazole in bulk and solid dosage form. The method was validated by using various parameters as per ICH guidelines.⁹

MATERIALS AND METHODS Materials

Ornidazole was obtained as a gift sample from Zydus Healthcare, Rangpo, East Sikkim. Ethanol and other reagents were of analytical grade. The U.V-Visible double beam spectrophotometer (Shimadzu-1700) with a fixed slit width (2 nm) and 10 millimeter quartz cell was used to obtain spectrum and absorbance measurement.

Methods

Procedure for Preparation of Calibration Curve

Ornidazole 10 mg was accurately weighed and diluted with 10 ml of ethanol. From this, 0.05ml, 0.1ml, 0.15ml, 0.2ml and 0.25ml were taken and diluted each

to 10 ml with ethanol to give solution of $5\mu g$, $10\mu g$, $15\mu g$, $20\mu g$ and $25\mu g$ concentration of drug per ml respectively. The solutions were scanned in U.V/Visible double beam spectrophotometer (Shimadzu-1700) against ethanol as a blank.

Procedure for Drug Recovery

In order to assess the drug recovery 10 tablets each from two market products (a) Orni- 500mg (Zydus Ornida (Aristo Healthcare) and (b) 500mg Pharmaceutical Pvt.Ltd) were weighed and powdered. From this, powder equivalent to 10mg was weighed, diluted to 10 ml with ethanol. The solution was shaken well and filtered through whatman filter paper (No.41). After suitable dilution, absorbance of final sample corresponding to 10 µg/ml was recorded against blank.To examine the absence of either positive or negative interference of excipients used in formulation, recovery studies were carried out at five different labels.

Table 1: Statistical Analysis (ANOVA Table) for Standard (Bulk) and Test Sample

Best-fit 95% confidence interval							
STANDARD LINEAR REGRESSION	Number of points =5	Parameter	Value	Error	From	То	
		Slope	0.03894	0.002250	0.03178	0.04610	
		Y-intercept	0.06830	0.03731	-0.05042	0.1870	
		X-intercept	-1.754				
		Correlation coefficient (r) = 0.9950 , r-squared = 0.9901					
		Standard deviation of residuals from line $(Sy.x) = 0.03557$					
		Source of Variation	Degrees of	Sum of	Mean		
			freedom	squares	square	F =	
		Linear regression	1	0.3791	0.3791	299.57	
		(Model)					
ANOVA TABLE FOR		Deviations from	3	0.003796	0.001265		
STANDARD		linearity (Residual)					
		Total	4	0.3829			
		The P value is 0.0004, considered extremely significant.					
TESTLINEAR REGRESSION	Number of points =5	Parameter	Value	Error	From	То	
		Slope	0.03806	0.002421	0.03036	0.04576	
		Y-intercept	0.2671	0.04015	0.1393	0.3949	
		X-intercept	-7.018				
		Correlation coefficient (r) = 0.9940 , r-squared = 0.9880					
		Standard deviation of residuals from line $(Sy.x) = 0.03828$					
		Source of Variation	Degrees of	Sum of	Mean		
			freedom	squares	square	F =	
ANOVA TABLE		Linear regression	1	0.3621	0.3621	247.08	
FOR TEST		(Model)					
		Deviations from	3	0.004397	0.001466		
		linearity (Residual)					
		Total	4	0.3665			
		The P value is 0.0006, considered extremely significant.					

PARAMETERS	STANDARD	TEST	
Absorption maximum(nm)	310.5nm	310.5nm	
Beer's law limit (µg/ml)	5-25 µg/ml	5-25 μg/ml	
Regression equation(y=mx+c)	y=0.038x+0.068	Y=0.038+0.267	
Slope (m)	0.038	0.038	
Intercept (c)	0.068	0.267	
Regression coefficient(r ²)	0.990	0.988	
Correlation coefficient(r)	0.9950	0.9940	
Standard deviation	0.0374	0.0211	
Molar absorptivity(mol ⁻¹ cm ⁻¹)	83457.5		
P value	0.0004	0.0006	
Linear regression	0.0389	0.0380	
Deviation from linearity	0.0355	0.0382	
Percentage Recovery \pm S.D		96.30-98±0.021%	

 Table 2 :Ornidazole Bulk and Test Parameters Comparative Study



------ 95% Confidence interval Figure 1: Linear Regression Curve for Standard Sample Ornidazole

RESULTS AND DISCUSSION

The Maximum absorption for pure ornidazole in ethanol using U.V spectrophotometer at 310.5nm was recorded. According to ICH guideline⁹ the method was validated. Linear regression of standard with correlation coefficient r= 0.9950 indicates a good linearity between absorbance and concentration range of 5-25 mcg/ml and linear regression both for standard and test sample of ornidazole described by figure 1 and 2. The statistical analysis for both standard (bulk) and test sample described by table 1 and the optical



----- 95% Confidence interval Figure 2: Linear Regression Curve for Test Sample Ornidazole

characters such as Beer's law limit, molar absorptivity and other parameters are summarized in table 2.

CONCLUSION

The proposed method obeyed beer's law in the concentration range of 5-25 μ g/ml with ethanol. The recovery studies were carried were found close to 100% indicating accuracy and precision of the proposed method. Hence this developed method could be used for routine estimation of ornidazole from tablet formulations.

ACKNOWLEDGEMENTS

Authors wish to give thanks to Calcutta Institute of Pharmaceutical Technology and A.H.S. authority for constant support and given research laboratory to carry out this project work. We also have given a special thanks to Zydus Healthcare, Rangpo, East Sikkim, India for providing gift sample of ornidazole. We also acknowledge the help provided by our fellow colleagues in completion of the project.

REFERENCES

1. Mubeen, G, Prakash, V, Somashekar, PL, Uvesh, K. Spectrophotometric method for determination of ornidazole. Inter. J. Chem. Tech. Researc. 2009; 1(2): 318-321.

2. Singh, P, Mittal, R, Sharma, GC, Singh, S, Singh, A. Ornidazole: Comprehensive profile, In: Profiles of drug substances excipients and related methodology. Elsevier Inc.; 2003: 30(123), 125-173.

3. Merdjan, H, Bonnat, C, Singlas, E, Diquite, B, Measurement of ornidazole by high performance liquid chromatography. J. Chromatogr. 1983; 273(2): 475-480.

4. Groppi, A, Papa, P, Montaqna, M, Carosi, G. Determination of ornidazole in human plasma and red blood cells using high performance liquid chromatography. J. Chromatogr. 1986; 380(2): 437-442.

 Heizmann, P, Geschke, R, Zinapold K. Determination of ornidazole and its metabolites in biological fluid. J. Chromtogr. 1990; 534: 233-240.
 Oexkan, SA, Senturk, Z, Biryol, I. Determination of ornidazole in pharmaceutical dosage forms based on reduction at an activated glassy carbon electrode. Inter. J. Pharm. 1997; 157: 137-144.

7. Salman, A, Sumer, C. Spectrophotometric assay of ornidazole and nimorazole. Scient. Pharmaceuti. 1996; 64(2): 145-149.

8. Chakraborty, M, Gupta, BK, Debnath, R, Pal, RN, Mazumder, R, Maity, T, Sinha, DK. A new spectrophotometric method development for determination of rimonabant in bulk and tablets Inter. J. Phama. Pharmceut. Sci. 2009; 1(Suppl.1): 162-164.

9. I.C.H. Harmonized Tripartite guideline. Recommended for adaptation at step-4 of the ICH Process. By ICH steering committee, 1996.
