



Synthesis of 3-[4-Amino-N-substituted-o-anisamido-5-yl]amino-5-substituted-1,2,4-dithiazole

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Abstract : A novel series of 3-{4-Amino-N-[2-(dithethylamino)ethyl]-o-anisamido-5-yl}amino-5-substitutedimino-1,2,4-dithiazole was synthesized by the oxidative cyclization of 4-amino-5-substituteddithiobiureto-N-[2-(dithethylamino)ethyl]-o-anisamide in chloroform medium by making the use of liquid bromine as oxidizing agent. The products were characterized and justified on the basis of elemental analysis, chemical characteristics and spectral studies.

Keywords : 3-{4-Amino-N-[2-(dithethylamino)ethyl]-o-anisamido-5-yl}amino-5-substitutedimino-1,2,4-dithiazole, 4-amino-5-substituteddithiobiureto-N-[2-(dithethylamino)ethyl]-o-anisamide, Bromine in Chloroform.

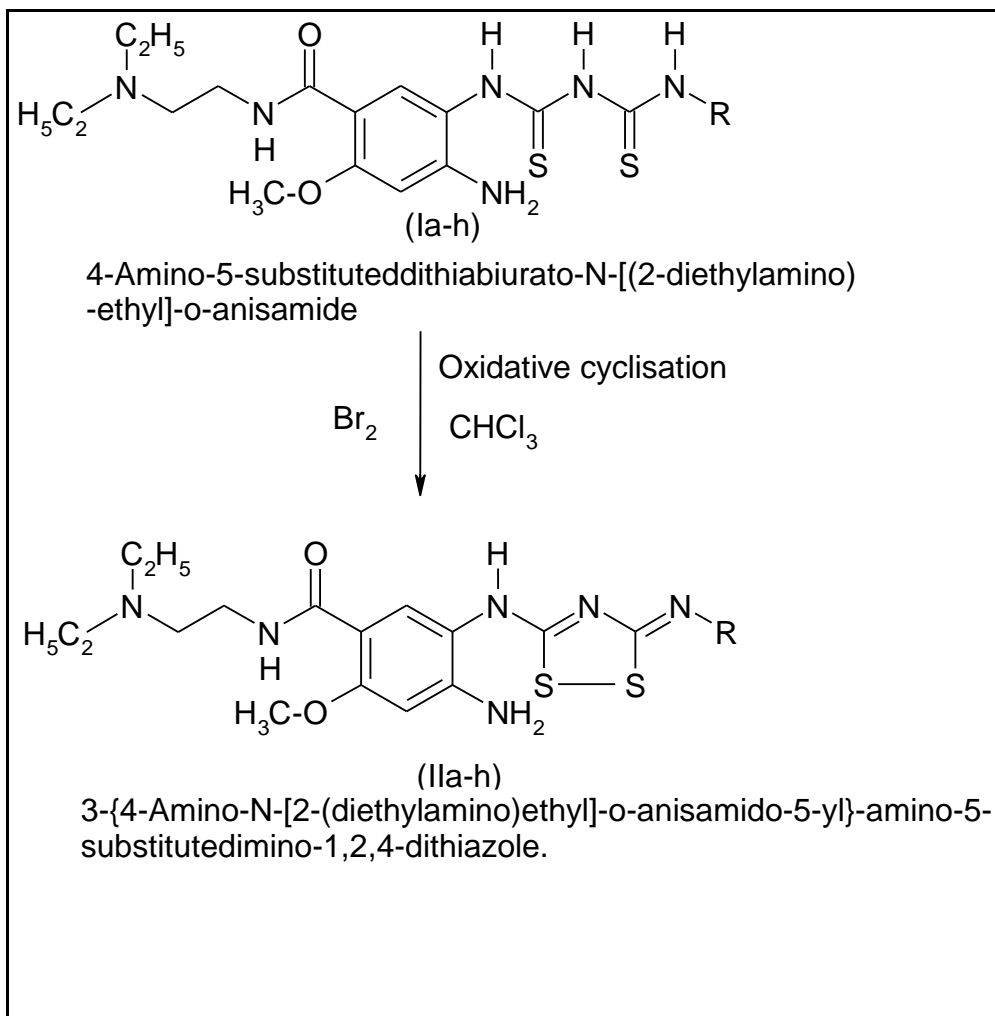
Introduction:

The literature survey shows that the dithiazolo, dithiazino and thiadiazino nucleus containing compounds are widely used in medicinal, industrial, agricultural, biochemical sciences^{1,7}. It was considered that these nucleus containing drugs possess, antibacterial⁸, antidiabetic⁹, amoebicidal¹⁰ herbicidal¹¹ properties. Iminosubstituted thiadiazolo nucleus also possesses noticeable pharmaceutical activities.

The heterocyclic compounds containing nitrogen, nitrogen and sulphur have gained huge importance in our life. It was shown that thiadiazoles are also effective against copper corrosion¹² and additive in lubricating oil¹³.

In recent years several 1,2,4-dithiazole and their derivatives were found to have prominent pharmacological activities such as anticonvulsant, analgesic anti-inflammatory activity. Dobolkar and Ansari¹⁴ had successfully investigated oxidative cyclisation of cyanoamidino substituted thiocarbamide and N-substitutedformidinothiocarbamides. Various researchers¹⁵⁻¹⁹ studied oxidative cyclisation of 1,3,4-thiadiazoles, 1,3,4-thiadiazolines and 1,2,4-triazoles.

As a part of research work presently undertaken in this laboratory in the synthesis of heterocycles and heterocycles, it was thought interesting to investigate the cyclisation of 4-amino-5-substituteddithiobiureto-N-[2-(diethylamino)ethyl]-o-anisamides with liquid bromine in chloroform medium to obtain a novel series of 3-{4-amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5-substitutedimino-1,2,4-dithiazoles which are hitherto unknown. The present work describes a suitable, convenient and somewhat direct method for this synthesis and depicted below



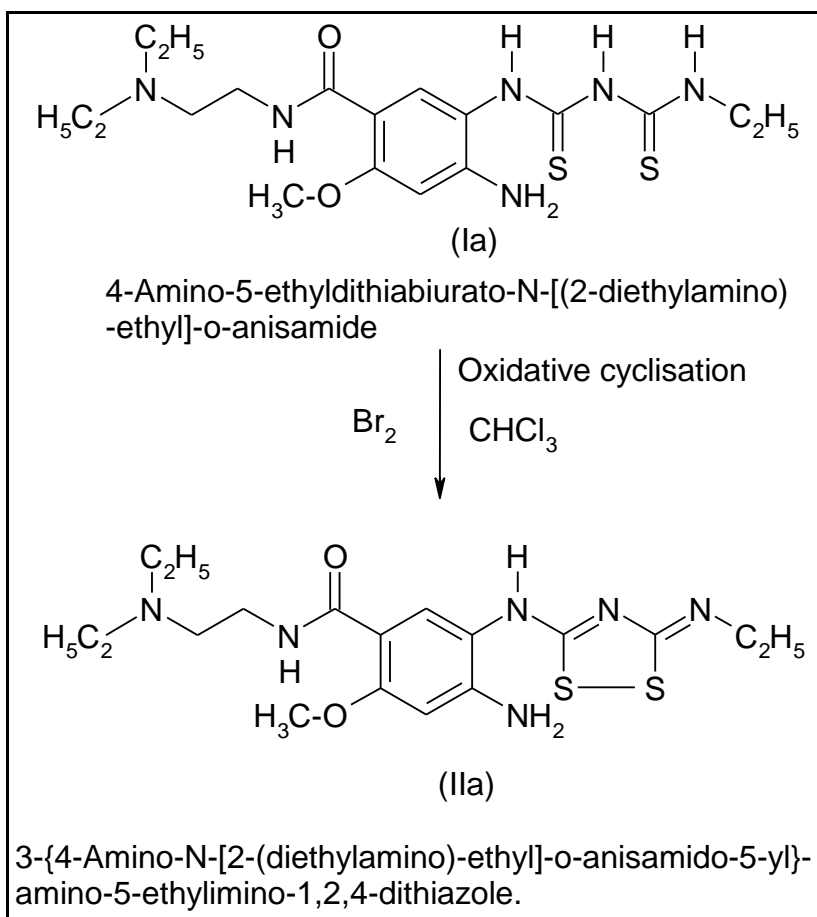
R= , methyl, t-butyl, phenyl, p-chlorophenyl, ethyl, o-tolyl, m-tolyl, p-tolyl.

Experimental:

The melting point of the all synthesized compounds was recorded using hot paraffin bath. The carbon and hydrogen analysis were carried out on Carlo-Ebra 1106 analyzer. Nitrogen estimation was carried out on Colman-N-analyzer-29. IR spectra were recorded on Perkin Elmer Spectrometer in range $4000-400\text{ cm}^{-1}$ in KBr pellets. PMR spectra were recorded on Bruker 400F spectrometer with TMS as internal standard using CDCl_3 and DMSO-d_6 as solvent. The purity of compound was checked on silica Gel-G Pellets by TLC with layer thickness of 0.3 mm. All chemicals used were AR-grade.

Synthesis of 3-{4-Amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5-ethylimino-1,2,4-dithiazole

3-{4-Amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5-ethylimino-1,2,4-dithiazoles was synthesized by the oxidative cyclization of 4-amino-5-ethyl dithiobiureto-N-[2-(diethylamino)ethyl]-o-anisamide with liquid bromine in presence of chloroform. A paste of 4-amino-5-ethyl dithiobiureto-N-[2-(diethylamino)ethyl]-o-anisamide was prepared in chloroform to it 10% liquid bromine in chloroform was added with constant stirring. Initially the colors of bromine disappear in the reaction mixture. The reaction mixture was allowed to stand for 8 hours, it afforded brown colored product. It was recrystallised from ethanol, yield 90%, m.p. 251°C .



Propertise :

It is brown colour crystalline solid having melting point 251°C. It gave positive test gave for nitrogen and sulphur. It was desulphurized by alkaline plumbite solution. It formed picrate having melting point 179°C. **Elemental Analysis** :C[(found 50.03%) calculated 50.94], H[(found 06.20%) calculated 06.60], N[(found 19.00%) calculated 19.81], S[(found 14.50%) calculated 15.09]. **IR Spectrum**: The IR spectrum was carried out in KBr pellets : 3429.00 (N-H stretching), 2923.24 (Ar-C=C stretching), 1639.05 (N-C=O stretching), 1506.08 (C=N stretching), 1092.21 (C-N stretching), 774.31 (C-S stretching). **NMR Spectrum**: The NMR spectrum of compound was carried out in CDCl₃ and DMSO-d₆. This spectrum distinctly displayed the signals due to Ar-H protons at δ 7.3000-7.152 ppm, -NH proton at δ 4.9072-4.2568 ppm, NH₂ protons at δ 3.9058-3.2062 ppm, -OCH₃ protons at δ 3.2009-3.0183 ppm, CH₂ protons at δ 2.9955-2.0929 ppm and N-CH₃ protons at δ 1.2269-1.1268 ppm.

From the above properties and spectral analysis of the compound (IIa) was assigned the structure as 3-{4-amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5-ethylimino-1,2,4-dithiazoles (IIa)

Similarly, 4-amino-5-methylthiobiureto-N-[2-(diethylamino)ethyl]-o-anisamide (Ib), 4-amino-5-t-butylthiobiureto-N-[2-(diethylamino)ethyl]-o-anisamide (Ic), 4-amino-5-p-chloro-phenylthiobiureto-N-[2-(diethylamino)ethyl]-o-anisamide (Id), 4-amino-5-o-tolyldithiobiureto-N-[2-(diethylamino)ethyl]-o-anisamide (Ie), 4-amino-5-m-tolyldithiobiureto-N-[2-(diethylamino)ethyl]-o-anisamide (If), 4-amino-5-p-tolyldithiobiureto-N-[2-(diethylamino)ethyl]-o-anisamide (Ig), were successfully oxidative cyclised with bromine in chloroform medium to isolate 3-{4-amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5-methylimino-1,2,4-dithiazole (IIb), 3-{4-amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5-t-butylimino-1,2,4-dithiazole (IIc), 3-{4-amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5-o-chlorophenylimino-1,2,4-dithiazole (IId), 3-{4-amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5-o-tolylimino-1,2,4-dithiazole (IIe), 3-{4-amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5-m-tolylimino-1,2,4-dithiazole (IIIf), 3-{4-amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5-p-tolylimino-1,2,4-dithiazole (IIg) respectively by the above mention method in **Table No. 1**

Table No. 1

Sr. No	3-{4-Amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5- substituted imino-1,2,4-dithiazoles	Yield (%)	m.pt. (°C)
1.	3-{4-Amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5- methyl imino-1,2,4-dithiazole	88	245
2.	3-{4-Amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5- t-butyl imino-1,2,4-dithiazole	81	249
3.	3-{4-Amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5- p-chlorophenyl imino-1,2,4-dithiazole	85	265
4.	3-{4-Amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5- o-tolyl imino-1,2,4-dithiazole	90	261
5.	3-{4-Amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5- m-tolyl imino-1,2,4-dithiazole	80	255
6.	3-{4-Amino-N-[2-(diethylamino)ethyl]-o-anisamido-5-yl}-amino-5- p-tolyl imino-1,2,4-dithiazole	90	270

References:

- Bhattacharya A, Singh T, Verma VK. 1,3,4- thiadiazoles as potential EP additives—a tribological evaluation using a four-ball test, *Tribol. Int.*, 1995, 28(3); 189.
- Ghosh SK. *Advanced Organic Chemistry*, 2nd Ed., Calcatta, 1998, (a) P-410,(b)P-412.
- Lapman G, Pavia D. and Kriz G. *Introduction to Spectroscopy*, Asia a Pta Ltd., 3rd Ed., Singapor, 2004, (a) P-68-69,(b) P-43.
- TayadeDT. *Asian Jr. of Chemistry*, 1995, 7(4); 890-891.
- Hamady NA, Abdel-Aziz HA, Farog AM, Fakhri Issa. MA. *Monatshefte fur chemie*, 2007,138; 1001-1010.
- Patel BV, Patel HS, Patel KC. *Ind. J. Chem, B*, 2008, 47B (06); 0376-4699.
- Ali T. El-sayad and Ibrahim MA. *J.Braz. Chem. Soc.*, 2010,446(2117); 1445-1468.
- Chan-Thaw CE, Villa A, Katekonon P, Sus D, Thomas A. and Prate L. *NanoLett*, 2010, 10(2); 537-541.
- Kurumurthy C, Veeraswamy B. Rao PS, Kumar GS, Reddy VL, RaoJV,Narsaiah B. *Bioorganic and Medicinal Chemistry Lett*, 2014, 24(3); 746-749.
- Sztanke K, Pasternak K, Rajtar B, Sztakne M, Majek M. and Polz Dacewicz M.J. *Bioorganic and Med.Chem*, 2007, 15; 5480-5486.
- Simanek EE, Abdou H, Lalwani S, Lim JJ, Mintzer M, VendittoVJ,Vittur B., *Proc. R. Soc. A*, 2003, 466(2117); 1445-1468.
- Krutz LJ, ShanerDL,Weaver MA, Webb R.M.T Zablutowicz RM, Reddy KN, Huang Y. and Thomson SJ, *Pest Management Sci.*,2010,66(5); 461-481.
- Sherif EM, Park SM,*ElectrochimicaActa*,2006,51; 6556-6562.
- Dobolkar VV, Ansari FY.*Acta Poloniac Pharmacelica-Drug research*, 2008, 65(5); 521-526.
- Lim J, Mintzer MA, Perez LM,Simanek EE, *Org.Lett.*,2010, 12(6); 1148-1151.
- MiranoK,*Chem. Abstr.*,1973, 79; 137200.
- Zhuo J, He C, Yao W, United States, Patent Application Publication,US2013/0345224 A1, 2013.
- Pittis WJ, Guo J, DharTG,Shen Z, Gu H, Watterson SH, Bednarz MS, *J.Bioorg.Med.Chem.*, 12(2), 19.
- Hajiduk PJ, Dinges J, Schkeryantz JM, Janowick D, Kaminski M, Tufano M, Augeri.J.*Med. Chem.*,1999, 42; 3852-3859.
