

Antidiabetic Activity of ARA (*Ficus Racemosa*) from ACEH

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Abstract : Isolation of chemical compounds from hexane extract of the stem bark of *Ficus racemosa* (fig), starting by maceration of 3 kg of bark *F.racemosa*, and obtained 68 g of extract concentrated, then separated by gravity column to obtained 4 groups of fraction (A, B, C, and D). Separation of the group fraction A was obtained isolates A4, allegedly α -Amyrin dodecanoic, which is based on spectroscopy by, FTIR, ¹H-NMR, and ¹³C-NMR, and also, mass-spectrometry. Antidiabetic activity by glucose tolerance in mice. Activity lowers blood sugar in mice at 30th minutes, sorted greatest to the smallest is fraction group D (200.67 mg/dL), the crude extract (99.00 mg/dL), fraction group of B (77.67 mg/dL), fraction group C (71.33 mg/dL), fraction group A (28.67 mg/dL) and isolates A4 (4.67 mg/dL). Activity lowers blood sugar in mice at 60th minutes, sorted greatest to the smallest is fraction group D (136.33 mg/dL), the crude extract (51.33 mg/dL), fraction group of B (12.00 mg/dL), while isolates A4 to raise blood glucose levels of 8.67 mg/dL, as well as fraction group of B and group fraction C, to raise the blood glucose respectively 39.00 mg/dL and 17.67 mg/dL. Activity lowers blood sugar in mice at 90th minutes, sorted greatest to the smallest, is the fraction group of D (46.33 mg/dL), isolates A4 (43.00 mg/dL), the group of fraction A (19.67 mg/dL), the fraction C group (6.67 mg/dL) and crude extract (4.33 mg/dL). Group fraction B, at the 90th minute, to raise the blood glucose levels of mice of 36.33 mg/dL. Activity lowers blood sugar in mice at 120th minutes, sorted greatest to the smallest, respectively from fraction group of the D (44.33 mg / dL), group C fraction (29.33 mg/dL), group A fraction (25.67 mg/dL), fraction group B (12.67 mg/dL) and crude extract (3.00 mg/dL). Group D bark fraction *F.racemosa* more active than others at 30 minutes of treatment, the level of 95%, ($p < 0.05$).

Keywords : *Ficus racemosa*, Antidiabetic, α -Amyrin dodecanoic, glucose tolerance.

Introduction

Diabetes is a global health problem, at least 171 million people worldwide suffer from diabetes, which is characterized by a chronic metabolic disorder with a high concentration of glucose in the blood [1]. This disease affects the nervous system of the eye, narrowing of blood vessels [2], insulin resistance, obesity, hypertension, and dyslipidemia [3].

Drug needs are great, and sometimes drugs individualistic, so searching for drugs continues to be done, and one of the plants is *Ficus racemosa*. This plant is widely spread widely throughout northern India and Australia [4]. In India, this plant has long been used for the prevention, treatment, and cure of various diseases

such as obesity and diabetes, jaundice, dysentery, diarrhea, inflammation [5], skin diseases, ulcers, asthma, gonorrhoea, menorrhagia, vaginal discharge, hemoptysis, urine disease and constipation [6].

Ficus racemosa plant contains many alkaloids, triterpenoids, steroids and flavonoids [7]. The roots of this plant contain long-chain hydrocarbon compounds, compounds triterpene lanostane, isocoumarin and compounds phytosteroid. Fruit, sap, leaves, wood and bark of *F. racemosa* the plant contains compounds triterpenoids, steroids, coumarin and phenolic esters [6].

The results showed that the fraction and β -Amyrin acetate from ethyl acetate extract at a dose of 1.5 g / kg can decrease by 2.89% and 2.31% weight of mice, respectively [8].

Material and Method

Plant Material

The bark of *Ficus racemosa* plants (ara) was taken in the Sarah, Aceh Besar, Banda Aceh in 2015. The plant was identified at Departement of Biology, University of Sumatera Utara, Medanense, Medan.

Spectroscopic Investigation

Mass spectra were measured using a Shimadzu GC-MS QP 2010 Ultra. 1D spectrum, ¹H NMR measured in CDCl₃ solvents, by JEOL 400 MHz spectrophotometer and spectrum ¹³C NMR measured in CDCl₃ solvents by a JEOL 125 MHz spectrophotometer. Column chromatography was conducted on silica gel 60 (70-230 mesh Merck). TLC analysis was Carried out by using precoated silica gel plates (Merck).

Testing Phytochemicals

The method used for testing of phytochemical can be found in Phytochemical methods, Simplified Determination Method to Analyze plant [9]

Extraction And Isolation Of Terpenoids (α -Amyrin Acetate) From The Bark Of *Ficus Racemosa*

The bark of *Ficus racemosa* 3 kg that has dried, were macerated with n-hexane for 3 x 24 hours, then filtered, the filtrate obtained, then evaporated with a rotary evaporator, and the hexane extract obtained as much as 68.50 grams or 2.28%. N-hexane extract was tested its antidiabetic activity to male Swiss Webster mice and characterized with GC-MS.

Fractionation Of Extract Hexane

A total of 35 grams of concentrated n-hexane extract was fractionated using column chromatography gravity. Eluent system is done using a gradient elution with n-hexane: ethyl acetate at a ratio of 100: 0; 95: 0.5; 90:10; 80:20; 70:30 and 50:50. The result of the separation of n-hexane extract obtained by 135 fractions, which consists of four groups of fractions, namely fraction group A (44-45, 13:03 g); fraction B group (46-48, 6.3 g); fraction group C (49-52, 2.17 g); fraction group D (53-135, 5.92 g). Fraction group of A, shaped yellow solid, as much as 13.03 gram (most clean than other fractions), was rechromatographed using column chromatography gravity and obtained isolates A4. Furthermore, A4 isolates characterized using, FTIR, MS, ¹H NMR and ¹³C NMR.

Glucose Tolerance Test[10]

Before use, the mice were acclimatized for 7 days in laboratory conditions as well as getting enough food and drinks. After 7 days, selected mice were healthy, characterized by weight stable or increased and did not show any abnormal behavior. Mice were divided into 8 groups, each of the groups contain three of mice. Group I: diabetic control was given CMC-Na (1%), group II: the standard drug glibenclamide was given orally at dose of 0.45 mg/kg BW, group III: treated with 50 mg/kg BW (effective dose) of fraction group of A (contain A4 isolates), group IV: treated with 50 mg/kg BW, hexane extract of *Ficus racemosa*. group V: treated with 50 mg/kg BW, of fraction group of B. Group VI: treated with 50 mg/kg BW, of fraction group

of C, group VII: treated with 50 mg/kg of fraction group of D, group VIII: treated with 50 mg/kg BW, isolate A4 [11]. The extract and the fraction group A, B, C, D, isolates were suspended with CMC-Na 1%.

Having fasted for 20-24 hours, the weight of mice were weighed, fasting blood glucose levels were measured and given treatment (above). After 30 minutes later, the entire group was given a dose of 3 g glucose/kg BW orally. Furthermore fasting blood glucose levels were recorded at 30th, 60th, 90th, and 120th minutes after glucose loading.

Samples Blood

Mice were put in a box modifications (restrainer), tail cleaned with a wet cotton so that the dirt is gone, then smeared with alcohol 70% v/v. Blood was drawn from the lateral tail vein, which was cut aseptically approximately 1-2 mm from the tip of the tail without anesthesia, blood droplets first removed, then the next drop of blood dripped on the strip One Touch Horizon.

Statistical Analysis

Statistical analysis was performed using Statistical Product And Service Solution (SPSS) Program. Analysis of variance was performed using ANOVA one way Post hoc analysis procedures, significant differences ($p < 0.05$) using Tukey[12]

Results and Discussion

Phytochemical test results from the bark of fresh *F.racemosa* and hexane extract of *F.racemosa* indicate a class of secondary metabolites each, steroids and terpenoids. Test results phytochemical from fractions group A, B, C, and D showed that all of these fractions group contain secondary metabolites of terpenoids, which are marked with red color formation after administration with reagent Liebermann-Burchard. The isolates of A4 is also terpenoids.

Isolates A4 Characterization

Isolates A4, in the form of a white solid, from hexane extract. ¹H-NMR spectrum for proton aliphatic (CH₃, CH₂, and CH) on triterpenoid compounds shown in chemical shifts (δ H) 2 ppm, which is characteristic of proton cyclic of the basic framework of triterpenoids that are not separated properly [13]. Proton chemical shifts of 8 C-methyl at the chemical shift 0.92 ppm (s, 3H, C-23); 0.73 ppm (s, 3H, C-24); 0.79 ppm (s, 3H, C-25); 0.78 ppm (s, 3H, C-26); 1.12 ppm (s, 3H, C-27); 1.05 ppm (s, 3H, C-28); 0.87 ppm (s, 3H, C-29) and 0.84 ppm (s, 3H, C-30). One proton broad multiplet at 4.28 ppm (m, 1H) was ascribed to the protons of C-3, due to the influence of the electronegativity of O atoms and COR and presence of the olefin proton at δ H at 5.11 ppm (t, 1H), characteristics of the Δ^{12} (isolated double bonds, oleanane order) [14]. slider at 2.15 ppm showed ester groups. Chemical shift at 1.19 (m, 1H, C-20), shows the specific chemical shift to α -amyrin. Results ¹HNMR spectrum measured in CDCl₃ contained in Figure 1 below.

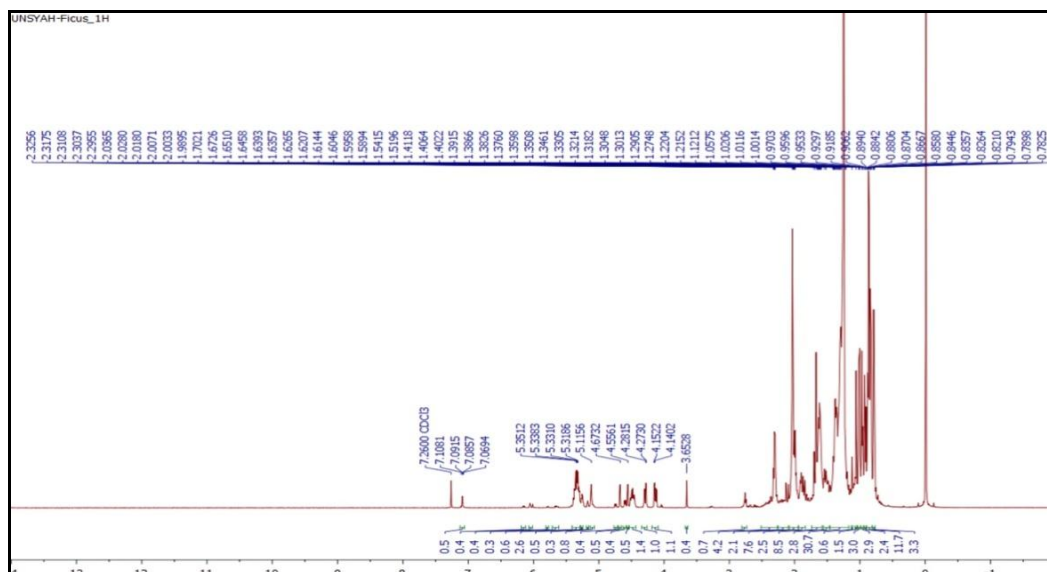


Figure 1. ¹H NMR spectrum of isolates A4

Based on the spectrum of Figure 2 below, shows 44 carbon atoms that have a cluster C = O with chemical shift δ is (173 ppm), ena (δ 139.59 and δ 124.29 ppm). The presence of carbon shift at 124.29 ppm (C-12) and 139.59 ppm (C-13) shows very similar characteristics to α -Amyrin, while the chemical shift of (C-12), 121.7-121.8 ppm and 145.2 ppm (C-13) on the β -Amyrin^{[15], [16]}. Eight C atom that binds a methyl group (C-methyl) of the compound α -Amyrin that the chemical shift 27.93ppm (C-23); 16.74ppm (C-24); 14.49 ppm(C-25); 16.86 ppm (C-26); 23.70ppm (C-27); 28.74 ppm (C-28); 23.36 ppm (C-29) and 21.40 ppm(C-30). Atom at C-12 and C-13 show chemical shifts are as high as 124.29 ppm and 139.59 ppm. The hallmark of the ¹³C-NMR spectrum of isolates A4 shows the order triterpenoids α -Amyrin^[17]. The results of the characterization of isolates A4 with a ¹³C-NMR instrument in Figure 2.

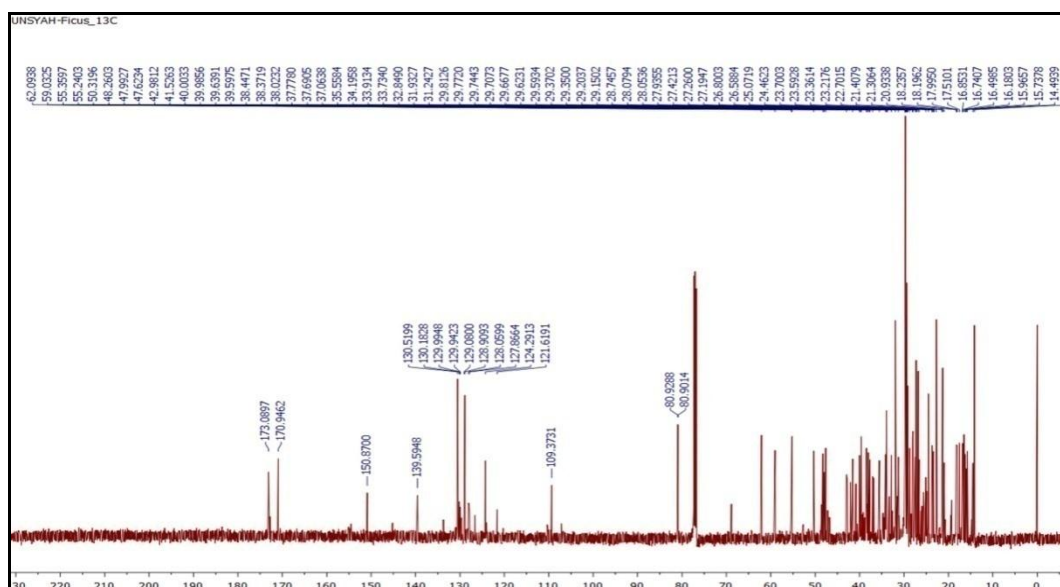


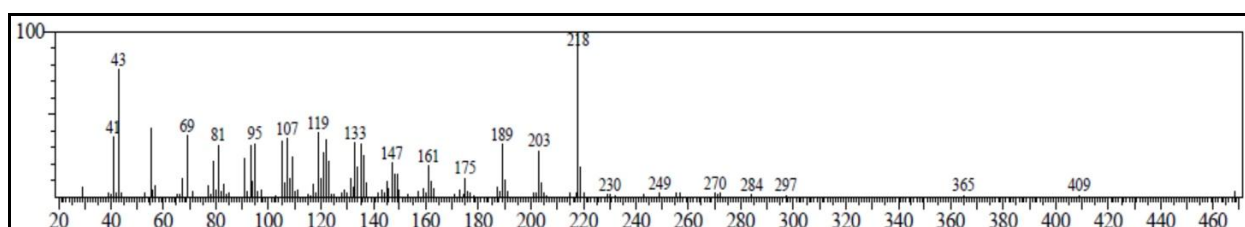
Figure 2. ¹³C NMR spectrum of isolates A4

Based on similarity chemical shift of carbon compounds A4 with compounds α -Amyrin, so isolates A4 compared with the standard α -Amyrin^[17]. Data comparison chemical shifts of ¹³CNMR isolates A4 with α -Amyrin standard can be seen in Table 1 below.

Table 1. Comparison of ¹H NMR spectral data and ¹³CNMR isolates A4 with the compound α -standard Amyrate acetate

Position H	Chemical shift (δ H) ppm (JHz)		Position C	Chemical shift (δ C) ppm (JHz)	
	A4	α - Amyrin acetate *		A4	α - Amyrin acetate *
1	1.27-1.35 (m)	1.28-1.35 (m)	1	38.44	38.61
2	1.67 (m)	1.67 (m)	2	24.46	24.10
3	4.28 (t)	4.43 (t)	3	80.90	79.41
4	-	-	4	38.02	38.01
5	0.84 (m)	0.84 (m)	5	55.35	55.72
6	1.54 (m)	1.54 (m)	6	18.23	19.03
7	1.38 (m)	1.38 (m)	7	23.70	23.86
8	-	-	8	41.52	41.10
9	1.54 (m)	1.54 (m)	9	48.26	48.15
10	-	-	10	37.69	37.43
11	1.62 (m)	1.62 (m)	11	18.23	18.20
12	5.11 (m)	5.12 (d)	12	124.29	125.04
13	-	-	13	139.59	140.10
14	-	-	14	42.98	43.40
15	0.86 (m)	0.86 (m)	15	29.15	29.12
16	0.79 (m)	0.79 (m)	16	37.69	37.43
17	-	-	17	33.73	33.21
18	1.98 (m)	1.91 (d)	18	59.03	59.53
19	1.00 (m)	1.00 (m)	19	40.00	40.25
20	1.98	1.98	20	38.44	38.97
21	1.30 (m)	1.30 (m)	21	31.24	31.68
22	-	-	22	41.52	40.97
23	0.92 (s)	0.93 (s)	23	27.93	27.91
24	0.77 (s)	0.73 (s)	24	16.74	16.78
25	0.79 (s)	0.80 (s)	25	14.49	14.89
26	0.78 (s)	0.77 (s)	26	16.85	16.90
27	1.12 (s)	1.10 (s)	27	23.70	23.81
28	1.05 (s)	1.05 (s)	28	28.74	28.53
29	0.87 (s)	0.87 (d)	29	23.36	23.46
30	0.84 (s)	0.84 (d)	30	21.40	21.52

* 16 Isolates A4 as α -Amyrin, reinforced by its data of MS, MS spectrum of isolates A4 in Figure 3.

**Figure 3. Mass spectrometry of the isolates A4**

In mass spectrometry, fragmentation patterns isolates A4 shows the molecular ion breakdown occurs at m/z 409, 218 (base peak), 203, and 189. Most of the triterpene compounds were found to have structural similarities with oleanane or Ursan compounds characterized by the base peak at m/z 218, the pattern of fragmentation the molecular ions are characteristic of the α -Amyrin compound, which has the same relative intensity at the peak of m/z 203 and 189 ^[16]. The fragmentation of m/z 409 is characteristic of α -amyrin

dodecanoic ^[16]. so that the isolates of A4 is a triterpene that binds ester groups. This is reinforced by spectrum Fourier Transform Infrared Spectroscopy (FT-IR). Spectrum FT-IR of A4 can be seen in Figure 4, absorbant carbonyl (C = O) of the ester at wave number 1735 cm⁻¹.

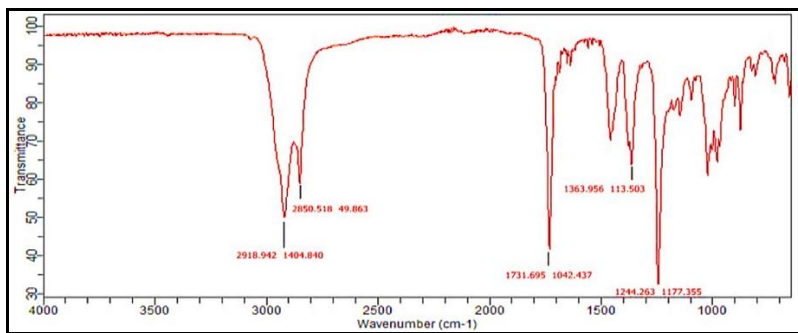


Figure 4. FT-IR spectra of isolates A4

Based on the results of the characterization of ¹H NMR and ¹³C NMR, MS, and FT-IR were allegedly A4 isolates is α -Amyrin dodecanoic. The structure of the isolates α -Amyrin dodecanoic can be seen in Figure 5.

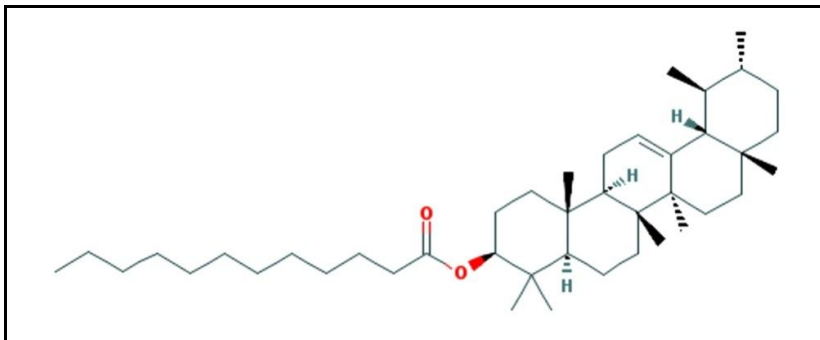


Figure 5. Structure of α -Amyrin dodecanoic

Antidiabetic (Hypoglycemia) Activity

Hypoglycemic activity of the extract, fractions, isolates A4, positive control, and negative control, in lowering blood glucose levels in mice can be seen in Figure 6 below

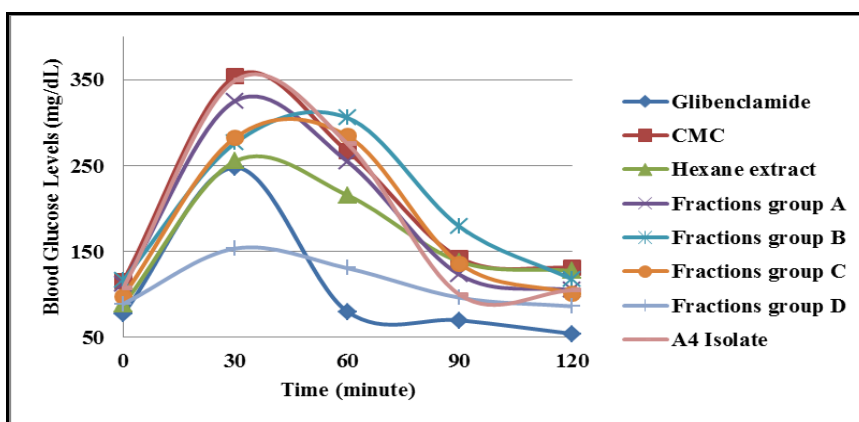


Figure 6. Graph decrease in blood glucose levels in mice against comparative glibenclamide and CMC 1%

Based on Figure 6 above blood glucose at 30 minutes after induction of glucose, all groups of mice experienced an increase in blood glucose levels. This increase is due to crude extract, fractions, A4, positive and negative controls is not yet control the sugar in the blood, so the blood sugar to rise.

Blood glucose in mice given the extract crude in the 30th minute increased levels of blood glucose to 255.33 mg/dL. At minute 60th, 90th and 120th experienced a decrease in blood glucose levels respectively to 215.67 mg / dL, 138.33 m /dL, and 127.67 mg/dL.

Mice were given of fraction A, at minute-30th experienced an increase in blood glucose levels into 325.67 mg/dL. At minute 60th, 90th and 120th experienced a decrease in blood glucose levels respectively to 255.00 mg / dL, 123.00 mg/dL, and 105.00 mg/dL. Mice were given a fraction B at minute 30th and 60th experienced a rise in blood glucose levels respectively to 276.67 mg/dL and 306.00 mg/dL. In the 90th minute, and 120th experienced a decrease in blood glucose levels respectively to 179.00 mg/dL and 118.00 mg/dL.

Groups of mice were given of fraction C at the 30th minute and 60th experienced an increase in blood glucose levels respectively to 283.00 mg/dL and 284,67 mg/dL. In the 90th minute, and 120th experienced a decrease in blood glucose levels respectively to 136.00 mg / dL and 101.33 mg/dL.

Groups of mice were given a fraction of D in the 30th minute, there is increased levels of blood glucose to 153.67 mg/dL. At minute 60th, 90th and 120th experienced a decrease in blood glucose levels respectively to 130.67 mg/dL, 96.33 mg / dL, and 86.33 mg/dL.

Groups of mice were given isolates A4 in the 30th minute, there is an increased level of blood glucose to 349.67 mg/dL. In the 60th minute, and 90th experienced a decrease in blood glucose levels respectively to 275.67 mg/dL, 99.67 mg/dL but in the 120th minute, there is an increased glucose level to 105.00 mg/dL.

Positive and negative control group at 30th minutes after treatment also showed a rise in blood glucose levels respectively to 248.00 mg/dL and 354.33 mg/dL. At 60th minutes after treatment showed a decrease in blood glucose levels respectively to 80.00 mg/dL and 267.00 mg/dL. Positive and negative control group at 90th minutes after treatment showed a decrease in blood glucose levels respectively to 70.00 mg/dL and 142.67 mg/dL and at 120th minutes after treatment showed a decrease in blood glucose levels respectively to 54.33 mg/dL and 130.67 mg/dL. Furthermore, the activity lowers blood sugar in mice is calculated by subtracting the blood glucose of the negative control mice, with the blood glucose of mice that given: crude extract, fraction group of A, fraction group of B, fraction group of C, fraction group of D and A4 isolates. Results of blood glucose reduction of the negative control group of rats with six kinds of the sample in Figure 7 below.

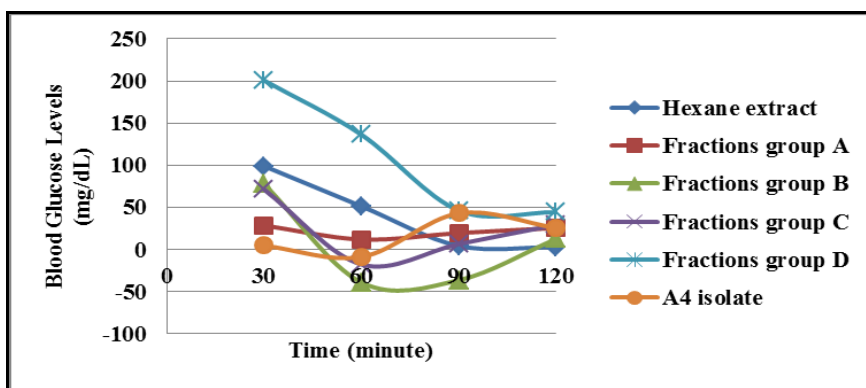


Figure 7. The result of a reduction in blood glucose in mice negative control group with blood glucose in rats that were given: crude extract, fraction group of A, fraction B group, fraction group C, group fraction D and isolates of the A4.

Based on Figure 7, it can be seen that the result of reduction in blood glucose levels of negative control with group given the sample at 30th minutes, sorted from the greatest to the smallest is fraction group D (200.67 mg/dL), hexane extract (99, 00 mg/dL), fraction group of B (77.67 mg/dL), group fraction C (71.33 mg/dL), the fraction group A (28.67 mg/dL) and isolates A4 (4.67 mg/dL).

At minute 60th, the result of a reduction in blood glucose negative control mice with mice that were given samples, sorted from the greatest to the least is as follows: group fraction D (136.33 mg / dL), the crude extract (51.33 mg/dL) group fraction A (12.00 mg/dL). Isolates A4, raise blood glucose levels as much as 8.67 mg/dL, and also fraction group of B and group fraction C, raise blood glucose respectively 39.00 mg / dL and 17.67 mg/dL.

At minute 90th, the result of a reduction in blood glucose negative control mice with mice that were given samples, sorted from the greatest to the least is as follows, the fraction D group (46.33 mg/dL), isolates A4 (43, 00 mg/dL), fraction group A (19.67 mg/dL), the fraction C group (6.67 mg/dL) and crude extract (4.33 mg/dL). Groups of mice were given a fraction B at 90th minutes to raise the blood glucose levels of 36.33 mg/dL.

At minute 120th, the result of a reduction in blood glucose negative control mice with mice that were given samples, sorted from the greatest to the least is as follows, the fraction D group (44.33 mg/dL), group C fraction (29.33 mg/dL), fraction group A (25.67 mg/dL), group fraction B (12.67 mg/dL) and crude extract (3.00 mg/dL).

Based on the above, generally all groups of fractions, extract n-hexane, and isolates A4, the bark of fig plants (*Ficus racemosa*), is relatively active in lowering blood glucose levels.

To see the difference in the activity of lowering of blood sugar in mice Webster male among the 6 types of samples above, (crude extract, group fraction A, group fraction B, group fraction C, fraction group of D, and compound of A4), is to use the Program Statistical Product And service Solution (SPSS), the analysis with ANOVA one-ways Post hoc analysis using Tukey, in order to obtain Table 2 below.

Table 2. Comparison of blood glucose levels decrease, crude extract, group fraction A, group fraction B, group fraction C, group fraction D, and isolates A4, with a concentration of 50 mg/Kg bb of mice, and the *p*-value (0.05) against glibenclamide (control positive) and CMC 1% (negative control)

Groups	Blood Glucose Levels (mg/dL)							
	30 minute	P	60 minute	P	90 minute	P	120 minute	P
Glibenclamide	248.00	-	80.00	-	70.00	-	54.33	-
		-		-		-		-
CMC 1%	354.33	-106.333	267.00	-	142.67	-72.667	130.67	-
		-		187.000		-		76.333*
Hexane extract	255.33	-7.333	215.67	-	138.33	-68.333	127.67	-
		99.000		135.667		4.333		73.333*
Fractions group A	325.67	-77.667	255.00	-	123.00	-53.000	105.00	-50.667
		28.667		175.000		19.667		25.667
Fractions group B	276.67	-28.667	306.00	-	179.00	-	118.00	-63.667
		77.667		226.000		109.000		12.667
Fractions group C	283.00	-35.000	284.67	-	136.00	-66.000	101.33	-47.000
		71.333		204.667		6.667		29.333
Fractions group D	153.67	94.333	130.67	-50.667	96.33	-26.333	86.33	-32.000
		200.667*		136.333		46.333		44.333
A ₄ isolate	349.67	-101.667	275.67	-	99.67	-29.667	105.00	-50.667
		4.667		195.667		43.000		25.667
				-8.667				

*=Significantly different from the control ($p < 0.05$), the first line of the positive control, line 2 to the negative control.

Group D fraction, significantly different from the negative control, and other samples at the time of 30 minutes

Conclusion

The test results antidiabetic n-hexane extract from the bark of fig plants (*Ficus racemosa*) have an activity to lower blood glucose levels in mice.

Group D bark fraction *F.racemosa* more active than others at 30 minutes of treatment, the level of 95%, ($p < 0.05$).

The results of the characterization of compounds A4 has similarities with the compound of α -Amyrin dodecanoic.

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