



**Thermodynamic and spectroscopic studies on complexes formation properties of Quercetin and Curcumin with Ni<sup>+2</sup> and pb<sup>+2</sup> and determination of stability constant by spectrophotometric method**

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**Abstract:**Quercetin and curcumin are are one of wide plenty dietary flavonoids. It has been scanned in the sharing of Pb(II),Ni(II) in Ethanol/H<sub>2</sub>O (40:60v/v)for Quercetin and Ethanol /H<sub>2</sub>O (60:40v/v)forcurcumin. The spectroscopic studies (UV-vis) were beneficial to consider the pertinent interaction of Quercetin and curcumin with Pb(II),Ni(II)ions .The chelation sites and dependence of the complex structure from the ligand /metal ratio. 1:1 (L:M)complex was indicated by Job's method of continuous variation.It was used to achieve the stoichiometric assembly of the complex.

**Keywords:**Thermodynamic, Quercetin and Curcumin, spectrophotometric.

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