# Crystal structure of 7-chloro-2,3,3a,4,9,9a-hexahydro-3,9,9-trimethyl-5-nitro-1H-cyclopenta[b]quinoline 

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#### Abstract

The title compound $\mathrm{C}_{16} \mathrm{H}_{19} \mathrm{Cl} \mathrm{N}_{2} \mathrm{O}_{2}$ is the product of reaction between 4chloroaniline and melonal in the presence of cupric nitrate and HCl . The product is the resultant of nitration of the aromatic ring and electrophilic aromatic cyclization. The heterocyclic ring at the center adopts a half-chair conformation and the five-membered ring has an envelope conformation. The crystal structure is stabilized by intra-molecular hydrogen bond. The molecular structure is stabilized by an intra-molecular $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bond, with an $S(6)$ ring motif.


Key words: crystal structure, quinolone, $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bond.

## Introduction

Tacrine is used to treat the symptoms of mild to moderate Alzheimer's disease and approved by United States Food and Drug Administration in $1993{ }^{1}$. One of the earliest and biggest changes the brain of Alzheimer's disease is that there is less of a chemical messenger called acetylcholine (ACh). Tacrine slows the breakdown of ACh, so it can build up and have a greater effect. It was the first centrally acting cholinesterase inhibitor approved for the treatment of Alzheimer's disease. The title compound is a congener of tacrine, has also been reported to be effective anti-alzheimeric agents ${ }^{2,3}$.

## Experimental

Crystals suitable for xray-diffraction studies were obtained by slow evaporation method. Data collection was carried-out using Oxford Diffraction Xcalibur Sapphire3 with graphite mono-chromatized Mo-K $\alpha$ radiation ( $\lambda=0.71703 \dot{A}$ ). The structure was solved by direct methods and refined on $\mathrm{F}^{2}$ by fullmatrix least squares procedures using the SHELXL programs ${ }^{3}$. The hydrogen atoms were identified using difference fourier. was used to create the image. For molecular graphics ORTEP- $3^{4}$ program and Mercury ${ }^{5}$ were used. The crystallographic data of the molecule is listed in Table-1.

Table: 1 Crystallographic data of the title compound

| Empirical formula | C15 H19 Cl N2 O2 |
| :---: | :---: |
| Formula weight | 294.77 |
| Temperature | 296(2) K |
| Wavelength | 0.71073 A |
| Crystal system, space group | P-1 |
| Unit cell dimensions | $\mathrm{a}=8.9847(13) \dot{\mathrm{A}} \quad \alpha=111.493(4)^{0}$ |
|  | $\mathrm{b}=9.1971(7) \dot{\mathrm{A}} \quad \beta=100.432(5)^{0}$ |
|  | $\mathrm{c}=9.8207(7) \dot{\mathrm{A}} \quad \gamma=97.188(5)^{0}$ |
| Volume | $726.17(13) \dot{\text { d }}^{3}$ |
| Z, Calculated density | 2, $1.348 \mathrm{Mg} / \mathrm{m}^{3}$ |
| Absorption coefficient | $0.266 \mathrm{~mm}^{-1}$ |
| F(000) | 312 |
| Crystal size | $0.19 \times 0.15 \times 0.11 \mathrm{~mm}$ |
| Theta range for data collection | 2.30 to $26.33^{0}$ |
| Limiting indices | $-11<=\mathrm{h}<=11,-11<=\mathrm{k}<=10,-10<=1<=12$ |
| Reflections collected / unique | $9988 / 2903$ [ R (int) $=0.0229]$ |
| $\begin{aligned} & \text { Completeness to theta }= \\ & 26.33 \end{aligned}$ | 98.2 \% |
| Max. and min. transmission | 0.9713 and 0.9512 |
| Refinement method | Full-matrix least-squares on $\mathrm{F}^{2}$ |
| Data / restraints / parameters | 2903 / 0 / 249 |
| Goodness-of-fit on $\mathrm{F}^{\wedge} 2$ | 1.048 |
| $\begin{array}{lcc} \hline \begin{array}{l} \text { Final } \\ {[I>2 \text { sigma(I) }]} \end{array} & \text { indices } \\ \hline \end{array}$ | R1 $=0.0376, \mathrm{wR} 2=0.0978$ |
| R indices (all data) | R1 $=0.0512$, wR2 $=0.1102$ |
| Largest diff. peak and hole | 0.244 and $-0.191 \mathrm{e} . \mathrm{A}^{-3}$ |

Synthesis of the compound


The quinoline derivative was prepared by the condensation 2,6-dimethyl-5- heptenaldehyde and 4chloro aniline in 1:1 molar ratio by refluxing in propan- 2 -ol and nitration of the resulting compound using $\mathrm{Cu}\left(\mathrm{NO}_{3}\right)_{2} .3 \mathrm{H}_{2} \mathrm{O}$ using $\mathrm{H}_{2} \mathrm{SO}_{4}$ as a catalyst. The mixture was left under reflux for 3 h . The solution was then left at room temperature. The solid product formed was separated by filtration, purified by crystallization
with ethanol washed with acetone and then dried in a vacuum over anhydrous calcium chloride (Vimal.et al., 2014). The brown colored product was formed in $86 \%$ yield.

## Results and Discussion

The title molecule has three fused rings consisting of two six- and one five-membered rings ( $\mathrm{A} / \mathrm{B} / \mathrm{C}$ ). The $\mathrm{A} / \mathrm{B}$ ring junction is trans-fused and $\mathrm{B} / \mathrm{C}$ is $c i s$-fused. The central ring B adopts a twist chair conformation with the puckering parameters $q 2=0.441(1)$ and $\varphi 2=193.9(3)$. The C8/C14/C23/C13/C ring has an envelope conformation, with C8 displaced from the other atoms (r.m.s. deviation $=0.026 \AA$ ) by 0.683 ( 6 ) $\AA$. The puckering are parameters. $q 2=0.422(1)$ and $\varphi 2=174.7(3)$. The packing is stabilized by weak intra-molecular N-H...O, C-H...O interactions. The N1-O2 $\cdots \mathrm{H} 2 \mathrm{~A}$ bond closes an $\mathrm{S}(5)$ ring motif. Table $1 \& 2$ gives the hydrogen bonding geometry and selected bond lengths, bond angles respectively Figures $1 \& 2$ give the ORTEP diagram and the packing of the molecules in the crystal cell respectively.

Table 1: Hydrogen bond geometry

| N_H...O | D-H $(\AA)$ | H...A $(\AA)$ | D---A $(\AA)$ | D-H...A $\left({ }^{\circ}\right)$ |
| :--- | :--- | :--- | :--- | :--- |
| N_H2A...O2 | 0.86 | 1.97 | $2.617(2)$ | 132 |

Table 2: Selected Bond lengths ( $\AA$ ) Selected bond angles ( ${ }^{\circ}$ )

| $\mathrm{Cl}(1)-\mathrm{C}(1)$ | $1.7435(17)$ | $\mathrm{O}(1)-\mathrm{N}(1)-\mathrm{O}(2)$ | $121.73(14)$ |
| :---: | :---: | :---: | :---: |
| $\mathrm{N}(1)-\mathrm{O}(1)$ | $1.2239(18)$ | $\mathrm{O}(1)-\mathrm{N}(1)-\mathrm{C}(3)$ | $118.05(15)$ |
| $\mathrm{N}(1)-\mathrm{O}(2)$ | $1.2294(18)$ | $\mathrm{O}(2)-\mathrm{N}(1)-\mathrm{C}(3)$ | $120.22(13)$ |
| $\mathrm{N}(1)-\mathrm{C}(3)$ | $1.448(2)$ | $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | $118.79(14)$ |
| $\mathrm{C}(2)-\mathrm{C}(1)$ | $1.355(2)$ | $\mathrm{C}(22)-\mathrm{N}(2)-\mathrm{C}(8)$ | $121.18(14)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)$ | $1.394(2)$ | $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(22)$ | $122.68(14)$ |
| $\mathrm{N}(2)-\mathrm{C}(22)$ | $1.352(2)$ | $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{N}(1)$ | $115.51(13)$ |
| $\mathrm{N}(2)-\mathrm{C}(8)$ | $1.442(2)$ | $\mathrm{C}(22)-\mathrm{C}(3)-\mathrm{N}(1)$ | $121.80(14)$ |
| $\mathrm{C}(3)-\mathrm{C}(22)$ | $1.417(2)$ | $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(1)$ | $122.05(15)$ |
| $\mathrm{C}(6)-\mathrm{C}(5)$ | $1.372(2)$ | $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(22)$ | $119.20(13)$ |
| $\mathrm{C}(6)-\mathrm{C}(1)$ | $1.393(2)$ | $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(10)$ | $119.72(14)$ |
| $\mathrm{C}(5)-\mathrm{C}(22)$ | $1.435(2)$ | $\mathrm{C}(22)-\mathrm{C}(5)-\mathrm{C}(10)$ | $121.08(14)$ |
| $\mathrm{C}(5)-\mathrm{C}(10)$ | $1.543(2)$ | $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(6)$ | $120.69(15)$ |
| $\mathrm{C}(10)-\mathrm{C}(9)$ | $1.532(2)$ | $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{Cl}(1)$ | $119.65(13)$ |
| $\mathrm{C}(10)-\mathrm{C}(12)$ | $1.536(2)$ | $\mathrm{C}(6)-\mathrm{C}(1)-\mathrm{Cl}(1)$ | $119.65(13)$ |
| $\mathrm{C}(10)-\mathrm{C}(11)$ | $1.537(2)$ | $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(12)$ | $108.76(15)$ |
| $\mathrm{C}(8)-\mathrm{C}(9)$ | $1.508(2)$ | $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)$ | $113.01(15)$ |
| $\mathrm{C}(8)-\mathrm{C}(14)$ | $1.523(2)$ | $\mathrm{C}(12)-\mathrm{C}(10)-\mathrm{C}(11)$ | $108.59(16)$ |
| $\mathrm{C}(14)-\mathrm{C}(15)$ | $1.508(3)$ | $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(5)$ | $106.74(12)$ |
| $\mathrm{C}(14)-\mathrm{C}(23)$ | $1.538(3)$ | $\mathrm{C}(12)-\mathrm{C}(10)-\mathrm{C}(5)$ | $110.35(14)$ |



Fig 1: ORTEP diagram of the molecule drawn at $30 \%$ probablility along with $S(6)$ motif


Fig 2. Crystal Packing viewed down ab-plane

## Results

The crystal structure of a novel quinoline was studied using single crystal X-ray diffraction method is reported The crystal structure is stabilized by C---H...O and Vander waals interactions.

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