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Investigation of Applying NF Membrane for Dye Removal Using Multivariate Regression Model

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Abstract: Recently, Nanofiltration plays an important role in dye wastewater treatment. The efficiency of membrane separation system for wastewater treatment plant is complicated due to different operating conditions involved in the separation process. Estimation of the Nanofiltration membrane performance is essential to evaluate the final water quality for further reuse. This work focus on the development of a simplified predictive model for membrane performance estimation under a set of parameters. A linear multiple regression model with coefficient of determination (\mathbb{R}^2) of 1 has been developed for COD values in the permeate stream, while (\mathbb{R}^2) of 0.85 for membrane rejection estimation. Seven investigated parameters have been related to formulate the model. The model has been applied for dye wastewater scheme comprising Nanofiltration with proper pretreatment. The model has been validated with an average deviation of 15% for membrane rejection and almost no deviation for COD level estimation in the permeate stream.

Keywords: Nanofiltration, Membrane separation, Wastewater Treatment, predictive model, Reactive dye.

1. Introduction

Textile dyeing industry is one of the largest water consuming industries. The effluent wastewater produced from the dyeing industries contains different chemicals and colouring agents and hence this effluent requires proper treatment before its discharge (Do & Chem¹). The dye house effluents are difficult to be treated because of the variable composition in the waste stream (Najmeh et.al.²). In addition to color, the textile wastewater the characteristics of high pH, high COD and low biodegradability (Mohammed³). Textile wastewater has a dramatic environmental impact and cannot be directly discharged to a receiving water body (Marrot & Roche⁴); therefore, textile wastewater treatment is mandatory due to the high water consumption rate and the environmental impacts (Chaia-Hung et.al.⁵).

The major sources of color in textile effluents are the dyes from the dyeing operations. There are three groups of textile dyes dependent on their state in solution as well as their charge: group N; neutral dyes such as disperse, vat and Sulphur dyes which are insoluble in water; group C; cationic dyes like basic dyes and group A; anionic dyes that is to acid, direct and reactive dyes which are soluble in water (Ahmad et.al.⁶). Furthermore, as a result of low biodegradability of most of the dye groups used in textile industry, biological treatment by activated sludge is not always effective due to aerobic biological treatment and oxidizing agent's resistance (Jiraratananon et.al.⁷). An advanced treatment technique is required, especially if the treated wastewater will be recycled/reused (Kapadan & Kargi⁸). Membrane separation is an optimal solution to remove color, COD and salinity (Aouni et.al.⁹).

Nanofiltration (NF), is a newly developed membrane technology for various water treatment and purification purposes (Koyuncu et.al.¹⁰); (Valentina et.al.¹¹). Although, membrane processes require proper pretreatment techniques, many approaches have been studied to minimize membrane fouling and reduce the costs (Koyuncu et.al.¹²). Pretreatment of feed water includes optimization of operating conditions such as pH as well as chemicals consumption and recovery ratio (Hong & Elinlech ¹³; Koyuncu et.al ¹⁴). Microfiltration (MF) and ultrafiltration (UF) can be used as a pretreatment step for Nanofiltration (Gozalvez-Zafrilla et.al.¹⁵; Debik et.al.¹⁶; Ong et.al.¹⁷; Barredo-Damas et.al.¹⁸; Alventosa-delara et.al.¹⁹).

Many studies have been carried out on color removal from textile wastewater by Nanofiltration processes. Lebrun et al., investigated electric field enhanced Nanofiltration process by BQ01 and NF45 membranes in dye removal. The results showed variations in dynamic permeability in the presence of electrolytes and according to the electrical potential applied (Lebrun et.al. ²⁰). Fuchs et al., studied membrane bioreactor (MBR) performance in textile wastewater treatment, and explored its ability to achieve water quality meeting reuse criteria. COD removal was found to vary between 60 and 95 % and COD levels reduced at lower volumetric loading rates that was tested (Fuchs et.al. ²¹). Das Gupta et al., studied Nanofiltration with MWCO = 400 Dalton to treat the effluent from a textile plant. Reactive black and red dyes were used and the membrane rejection was up to 92-94 % of the two dyes were achieved respectively, while COD removal was up to 94 % in cross-flow cell (Das Gupta et.al.²²).

De et.al, studied unstirred batch and cross flow Nanofiltration with MWCO = 400 Dalton to separate dye from aqueous solutions. Using the experimental results, and model parameters (i.e. the diffusivity of the solute (D) and real retention (R_r)) of the membrane were evaluated by optimizing the experimental flux and permeate concentration profiles (De et.al. ²³). Chaudhari et.al, investigated the color removal of commercially azo dyes under anaerobic conditions in wastewater. Color removal was achieved up to 99 % in both the dye containing reactors, whereas COD removals were up to 95 %, 92 % 94 % in control, orange and black dye containing reactors, respectively (Chaudhari et.al.²⁴).

Woei-Jye & Ismail²⁵, studied the performances of some commercial NF membranes and examined dye rejection, permeate flux and COD rejection as well as studying the transport mechanisms in NF membrane with a brief review of transport models of NF membrane. Jahangiri & Aminian²⁶, compare experimental results with neural network simulation in rejection estimation for two different nanofiltration membranes (NF-90 & DK-5). Arlindo & Isolina²⁷, studied the influence of osmotic pressure and solute adsorption on permeate flux during nanofiltration (NF) in dye wastewater treatment. He also predicts the rejection coefficient using the solution-diffusion model (Tahri et.al. ²⁸).

2. Methodology

The adopted methodology to come up with the performance indicators of applying nanofiltration membranes for dye wastewater is as follows:

- Collection and compiling of nanofiltration (NF) current practice in dye wastewater treatment.
- Determination of the governing process parameters.
- A mathematical correlation through multivariate regression to estimate the performance of using NF membrane in dye wastewater treatment.
- Validation of the regression model using published cases.

3. Results and Discussion

NF data for dye wastewater treatment have been reviewed and analyzed according to technical characteristics. Different data have been collected and excluded due to lack of total information needed regarding reactive dye separation using nanofiltration membranes. Table (1), presents compiled data regarding NF separation systems. The collected data have been refined to determine the parameters governing the separation process. Membranes used in determining this approach are NF90, NF 270, NF 200, SR90, DK2540F and 40-40 TS80 TSF. Multiple regression method is used to predict nanofiltration dye rejection and COD in the permeate values. The correlation developed are based on published data (Arlindo et.al.²⁷; Chia-Hung et.al.⁵;

Dye type	NF membrane Type	рН	Feed water conductivity (ms/cm)	Flux (l/m²h)	Transmembrane Pressure (bar)	Membrane pore size (Da)	COD _{in}	COD _{out}	Dye rejection	Ref
reactive dye	DK2540F	10.6	17	49	12	200	2450	612	98	[30]
reactive dye	DK2540F	9.2	14	40.5	12	200	2160	324	95	[31]
reactive dye	NF200	7.1	6	1.5	15	250	708	212	97	[32]
reactive dye	NF270	7.1	6	1.1	15	250	708	140	98	[32]
reactive dye	SR90	7.1	10	50	14	250	890	700	50	[33]
reactive dye	NF90	7.1	10	50	14	250	890	448	75	[33]
reactive dye	40-40 TS80 TSF	6.5	42	9	5	200	20	9	90	[34]

Table (1) Different NF membrane types characteristics for dye wastewater treatment

Hassani et.al.²⁹; Woei-Jye & Ismail ²⁵; Tahri et.al.²⁸; Najmeh et.al.²; Jahangiri & Aminian ²⁶; Ahmed et.al.⁶) assessed and assembled to formulate the regression model. Eight parameters were selected and investigated, six are considered as independent variables of the multivariate analysis, and two dependent variables. The independent variables are membrane flux, pH, dye rejection, trans-membrane pressure, membrane pore size, COD (in feed and permeate). Tables (2) and (3) shows selected compiled data used for model formulation, showing a strong relationship between the investigated variables and the estimated values.

The final forms of the formulated correlation are mentioned below in equation (1) and (2)

Dye Rejection (%) = (-880.248) + (104.552 * Cond) + (-25 * Flux) + (914.11 * Pore size) + (-56.218 * pH) + (24.686 * COD feed) (1)

COD rejection = 82655.46 + (5618.55 * pH) + (-912 * Feed Conductivity) + (154.8 * Flux) + (-404.63 * Pore size) + (-21.12 * Feed COD) + (-10.08 * Dye rejection).....(2)

Where:

Cond.= Feed conductivity (ms/cm)/10,

 $Flux = Flux (1/m^2h)/10,$

Pressure = Pressure (bar) /10,

COD feed = COD feed/100, and

Pore size = Pore size (Da)/200

Table (2) Coefficient of determination (R2) and the error Probability (P-value), Summary output for DyeRejection estimate using NF (Chia-Hung et.al.⁵; Hassani et.al.²⁹; Tahri et.al.²⁸, Ahmed et.al.⁶)

Regression Statistics									
Multiple R			0.91514						
R Square			0.857481						
Adjusted R Square			-0.97511						
Standard Error			17.67767						
Observations			7						
ANOVA									
	df	SS	MS	F					
Regression	6	1610.357	268.3929	1.030629					
Residual	1	312.5	312.5						
Total	7	1922.857							
	Coefficients	Standard Error	t Stat	P-value					
Intercept	-880.248	34379.25	-0.0256	0.983704					
рН	-56.2185	2188.507	-0.02569	0.98365					
Feed Conductivity	104.5524	3720.019	0.028105	0.982112					
Flux	-25	625	-0.04	0.974549					
Pressure	0	0	65535						
Pore size	914.1117	33370.39	0.027393						
COD feed	24.68629	854.8369	0.028878	0.981621					

The results showed that the rejection of reactive dyes by NF membranes highly dependent on the variation in the salt content of the feed stream and the membrane type (pore size) and independent on the applied pressure. These results are compatible with the published results of (Chollom et.al.³⁵). Salt rejections and permeate flux rates are dependent on feed pressure, however, for the studied range; no influence was noticed on the studied variables.

Table (3) Coefficient of determination (R2) and the error probability (P-value), Summary output forCOD value estimate in permeate by using NF

Regression Statistics								
Multiple R		1						
R Square		1						
Adjusted R Square		65535						
Standard Error		0						
Observations		7						
ANOVA								
	df	SS	MS					
Regression	8	380845.4286	47605.67857					
Residual	0	0	65535					
Total	8	380845.4286						
	Coefficients	Standard Error	t Stat					
Intercept	82655.45788	0	65535					
-	0	0	65535					
рН	5618.553443	0	65535					
Conductivity	-9120.01731	0	65535					
Flux	1548	0	65535					
Pressure 0		0	65535					
Pore size	-80926.39431	0	65535					
COD feed	-2112.941251	0	65535					
Membrane Rejection	-10.08	0	65535					

Figure (1), shows the deviation percent between estimated values and published case studies/data (Nur et.al.³⁶, Arlindo et.al.²⁷, Chia-Hung et.al.⁵, Hassani et.al.²⁹, Mehmet et.al.³⁷, Woei-Jye²⁵, Tahri et.al.²⁸, Aouni et.al.³⁰, Najmeh et.al.², Jahangiri & Aminian²⁶, Chollom et.al.³⁵, Reza et.al.³³, Najmeh et.al.³⁴, and Ahmed et.al.⁶). The values show that the percent deviation for COD value estimation is almost neglected, while for rejection estimation of about 15%.

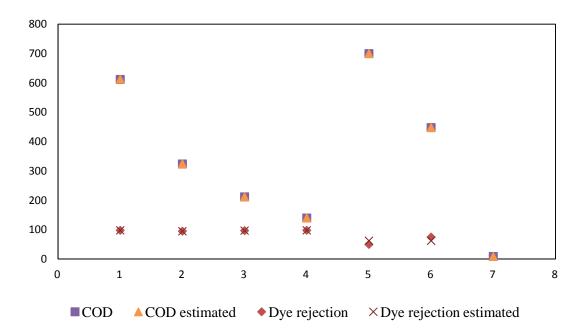


Figure (1) Comparison between estimated and published values

4. Conclusion

Due to the environmental concerns regarding treatment and recycling of dye wastewater. Nanofiltration presents a reliable option for dye wastewater treatment and presenting an option of water recycling. Multivariate regression models have been developed to predict membrane performance regarding reactive dyes rejection as well as COD concentration in the permeate stream. Eight parameters have been selected and investigated; six of them set were as the independent variables of the multivariate analysis and two dependent variables. Validation of the presented models by comparing the results with published data to give a reliable estimation on COD and rejection values of the membrane used under the pre-mentioned parameters.

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References

- 1. Do. J.S. and Chen. M.L., Deculturization of dyecont aining solutions by electro-coagulation, J. of Appl. Electrochemistry, 24, 785-790, 1990
- 2. Najmeh Askari, Mehrdad Farhadian, Amir Razmjou, Decolonization of ionic dyes from synthesized textile wastewater by nanofiltration using response surface methodology, Advances in Environmental Technology 2 (2012) 85-92.
- Mohammed Bassim Alqaragully," Removal of Textile Dyes (Maxilon Blue, and Methyl Orange) by Date Stones Activated Carbon ", International Journal of Advanced Research in Chemical Science (IJARCS) Volume 1, Issue 1, March 2014, PP 48-59, www.arcjournals.org
- 4. B. Marrot, N. Roche, Wastewater treatment and reuse in textile industries, a review, Res. Adv. Water Res. 3 (2002) 41–53.
- Chia-Hung liu, Jeng-Shiou Wu, Hsin-Chieh Chiu, Shing-Yi Suen, Khim Hoong Chu.Removal of anionic reactive dyes from water using anion exchange membranes as adsorbers. Water Research 41 (2007) 1491-1500.
- 6. Ahmad Akbari, Sandrine Desclaux, Jean-Christophe Rouch, Jean-Christophe Remigy, Application of nanofiltration hollow fiber membranes, developed by photo-grafting, to treatment of anionic dye solutions, HAL Id: hal-01290378 https://hal.archives-ouvertes.fr/hal-01290378, Submitted on 18 Mar 2016.

- 7. Jiraratananon, R., Sungpt, A., Luangsowan, P., 2000. Performance evaluations of nanofilteration membranes for treatment of effluents containing reactive dye and salt. Desalination 130, 177-183.
- 8. Kapdan, I.K., Kargi, F., 2002. Simultaneous biodegradation and adsorption of textile dyestuff in an activated sludge unit. Process Biochem. 37, 973-981.
- A. Aouni, C. Fersi, B. Cuartas-Uribe, A. Bes-Pia, M.I. Alcaina-Miranda, M. Dhahbi, Reactive dyes rejection and textile effluent treatment study using ultrafiltration and nanofiltration processes. Desalination 297 (2012) 87-96.
- 10. I. Koyuncu, E. Kural, D. Tupacik, Pilot scale nanofiltration membrane separation for waste management in textile industry. Water Sci. Technol. 43 (2001) 233-240.
- 11. Valentina Buscio 1, María García-Jiménez 1, Mercè Vilaseca 1, Victor López-Grimau 1,2, Martí Crespi 1 and Carmen Gutiérrez-Bouzán, Reuse of Textile Dyeing Effluents Treated with Coupled Nanofiltration and Electrochemical Processes, Materials 2016, 9, 490-502.
- 12. Koyuncu, I.; Topacik, D.; Wiesner, M. R., (2003). Factors influencing flux decline during nanofiltration of solutions containing dyes and salts. Water Res., 38(2), 432–440.
- 13. Hong, S.; Elimlech, M., (1997). Chemical and physical aspects of natural organic matter (NOM) fouling of nanofiltration membranes. J. Membr. Sci., 132(2), 159–181.
- Koyuncu, I., (2002). Reactive dye removal in dye/salt mixtures by nanofiltration membranes containing vinylsulphone dyes: Effects of feed concentration and cross flow velocity. Desalination, 143(1–3), 243– 253.
- 15. Gozálvez-Zafrilla, J.M.; Sanz-Escribano, D.; Lora-García, J.; León Hidalgo, M.C. Nanofiltration of secondary effluent for wastewater reuse in the textile industry. Desalination 2008, 222, 272–279.
- 16. Debik, E.; Kaykioglu, G.; Coban, A.; Koyuncu, I. Reuse of anaerobically and aerobically pre-treated textile wastewater by UF and NF membranes. Desalination 2010, 256, 174–180.
- 17. Ong, C.S.; Lau, W.J.; Ismail, F. Treatment of dyeing solution by NF membrane for decolorization and salt reduction. Desalin. Water Treat. 2012, 50, 245–253.
- Barredo-Damas, S.; Alcaina-Miranda, M.I.; Iborra-Clar, M.I.; Mendoza-Roca, J.A. Application of tubular ceramic ultrafiltration membranes for the treatment of integrated textile wastewaters. Chem. Eng. J. 2012, 192, 211–218.
- Alventosa-deLara, E.; Barredo-Damas, S.; Zuriaga-Agustí, E.; Alcaina-Miranda, M.I.; Iborra-Clar, M.I. Ultrafiltration ceramic membrane performance during the treatment of model solutions containing dye and salt. Sep. Purif. Technol. 2014, 129, 96–105.
- Lebrun, R.; Noel, I. M.; Bouchard, C. R., (2000). Electronanofiltration of a textile direct dye solution, Desalination, 129 (1-3), 125-136.
- 21. Fuchs, W.; Brike, M.; Schoeberl, P.; Chamam, B.; Braun, R., (2006). Advanced treatment of textile wastewater towards reuse using a membrane bioreactor, Process Biochemistry, 41 (8), 1751-1757.
- 22. Das Gupta, S.; Chakraborty, S.; Purkait, K. M.; De, S.; Basu, K. J., (2003). Nanofiltration of textile plant effluent for color removal and reduction in COD. Sep. Purif. Tech., 31 (2), 141-151.
- 23. De, S.; Chakraborty, S.; Bag, B. C.; Das Gupta, S.; Basu, K. J., (2004). Prediction of permeate flux and permeate concentration in nanofiltration of dye solution. Sep. Purif. Tech., 35 (2), 141-152.
- Chaudhari, S.; Manu, B., (2002). Anaerobic decolorisation of simulated textile wastewater containing azo dyes, Bioresource Tech., 82 (3), 225-231.
- 25. Woei-Jye L.; & Ismail A.F., (2009). Polymeric nanofiltration membranes for textile dye wastewater treatment: Preparation, performance evaluation, transport modeling, and fouling control a review. Desalination, 245:321.
- 26. Jahangiri, M.; & Aminian, A., (2012). Treatment of Textile Wastewater by Nanofiltration Membranes: A Neural Network Approach. Journal of Textile Science Engineering, 2:119.
- 27. Arlindo, C. G., Isolina, C. G. & Maria N., (2005). The role of adsorption on nanofiltration of azo dyes. Journal of Membrane Science, 255:157.
- Tahri, N., Ellouze, E., Jrad, A., & Amar R. B., (2010). Combination of Microfiltration and Nanofiltration processes applied to the reactive dyeing effluent treatment, Proceedings of the 3rd International CEMEPE & SECOTOX Conference Skiathos, June 19-24: 2011, ISBN 978-960-6865-43-5.
- 29. Hassani, A. H., Mirzayee, R., Nasseri, S., Borghei, M., Gholami, M., & Torabifar B., (2008). Nanofiltration process on dye removal from simulated textile wastewater. International Journal of Environmental Science and Technology, 5:3:401.
- A. Aouni , C. Fersi , B. Cuartas-Uribe , A. Bes-Pía , M.I. Alcaina-Miranda, M. Dhahbi, Reactive dyes rejection and textile effluent treatment study using ultrafiltration and nanofiltration processes, Desalination 297 (2012) 87-96.

- 31. Ahmed Mohamed Farid Shaaban, Azza I. Hafez, Mona A. Abdel-Fatah, Nabil Mahmoud Abdel-Monem, Mohamed Hanafy Mahmoud; Process engineering optimization of nanofiltration unit for the treatment of textile plant effluent in view of solution diffusion model, Egyptian Journal of Petroleum 25, 79–90, 2016. <u>http://dx.doi.org/10.1016/j.ejpe.2015.03.018</u>
- 32. MN Chollom, S Rathilal, VL Pillay and Dorcas Alfa, The applicability of nanofiltration for the treatment and reuse of textile reactive dye effluent, Water SA Vol. 41 No. 3 April 2015
- 33. H. Reza Rashidi, N. Meriam Nik Sulaiman, N. Awanis Hashim, C. Rosmani Che Hassan & M. Redzuan Ramli, Synthetic reactive dye wastewater treatment by using nano-membrane filtration, Desalination and Water Treatment, (2014) 1-10.
- 34. Najmeh, A.; Mehrdad F.; & Amir, R., (2015). Removal of textile dye from aqueous solutions by nanofiltration process. Iranian Journal of Environmental Technology, 1: 2: 43.
- 35. Chollom, M.N., Rathilal, S., Pillay V.L., & Dorcas A., (2015). The applicability of nanofiltration for the treatment and reuse of textile reactive dye effluent. Water SA, 41: 3 http://dx.doi.org/10.4314/wsa.v41i3.12
- 36. Nur, H. H.; Abdul-Wahab, M.; & Abdul-Amir, H. K., (2004). Nanofilteration of hazardous Congo red dye: Performance and flux decline analysis. Journal of Water Process Engineering, 4: 99.
- 37. Mehmet, C., Necati, K., & Ismail, K., (2008). Desalination of produced water from oil production fields by membrane processes. Desalination, 222:176.
