



International Journal of ChemTech Research

CODEN (USA): IJCRGG ISSN: 0974-4290 Vol.8, No.4, pp 2234-2240, 2015

The mass spectral fragmentation of substituted m-terphenyl and biarylderivatives

Periyasamy Murugan*

Department of Chemistry, Vel Tech University, Avadi , Chennai- 600 062, India.

Abstract: A series of m-terphenyl and biaryl derivatives have been investigated by EI mass spectrometry because of due to their versatile wide applications such as laser dyes, OLED, liquid crystals, solar cells and NLO properties. Mass spectral fragmentation of m-terphenyl and biaryl derivatives were thoroughly analyzed in this article. In general, all the compounds except **3b**, showed the molecular ion peak. Loss of CH₃ group from the Ar-CH₂CH₃ or N-CH₂CH₃ has been observed from their parent skeleton compounds **2a-g**. The most common loss of CN,NH₂,HCN,C₂H₄ groups were observed from the compounds**1a-e, 2a-g, 3a-c**, which are depicted in the **scheme (I,II,III & IV)**. The loss of CHO group is the characteristic of furfuryl moiety has been observed in compound **3c**(m/z 194).As a representative fragmentation pattern of compound **1a** depicted in **Scheme-V**.

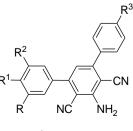
Key words: M-terphenyl, biaryl, OLED, NLO, furfuryl, HCN, CN, CHO.

Introduction

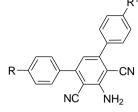
Mass spectrometry (MS) is unequivocally a powerful tool of characterization of organic compounds. In recent years, we have witnessed a significant development in the utilization of the terphenyls, particularly m-terphenyl¹ which are useful intermediate for building blocks for cyclophanes² to create a large molecular cavity³ and host-guest complexes⁴. The biaryl unit is represented in several types of compounds of current interest including natural products, polymers, advanced materials, liquid crystals, ligands and molecules of medicinal interest⁵. In continuation of our work on the synthesis of m-terphenyl and biaryl compounds having laser dyes⁶ and NLO properties⁷. In this paper, we would like to report the mass spectral fragmentation pattern of m-terphenyl and biaryl compounds **1 a-e , 2 a-g** and **3 a-c**. Similar loss of CH₃, HCHO and HCN groups were observed both methyl substituted compounds **1b** and **1d**.The characteristic loss of CHO group was observed in compound **3c**, it shows presence of furfuryl moiety.

Experimental

The mass spectra were recorded with Jeol-JMS-DX 303 HF and GCMS QP 5000 Shimadzu instruments. Synthesis of m-terphenyl and biaryl compounds **1a-e**, **2a-g**, **3a-c** were previously reported⁸ by our research group.

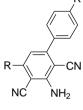


Compounds	R	R ¹	R^2	R ³
1a	Н	Н	Н	Н
1b	OCH₃	OCH₃	OCH₃	Н
1c	Н	Cl	Н	Н
1d	OCH₃	OCH₃	OCH₃	CH₃
1e	Н	Н	Н	CH ₃
			D1	



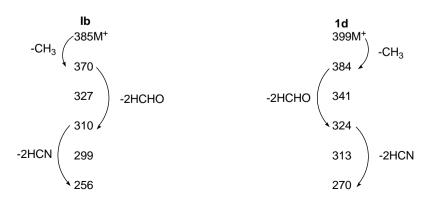
2a-g

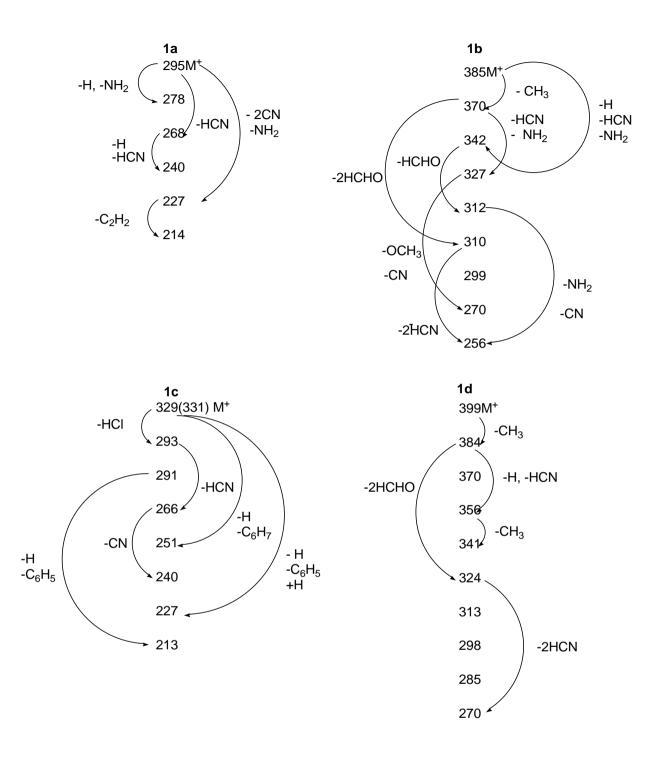
COMPOUND	R	R^1
2a	$N(CH_2CH_3)_2$	CH ₃
2b	$N(CH_2CH_3)_2$	CH_2CH_3
2c	N(CH ₃) ₃	CH ₂ CH ₃
2d	OCH ₃	CH ₃
2e	CH ₃	CH₃
2f	CH ₃	CH ₂ CH ₃
2g	Н	CH ₂ CH ₃



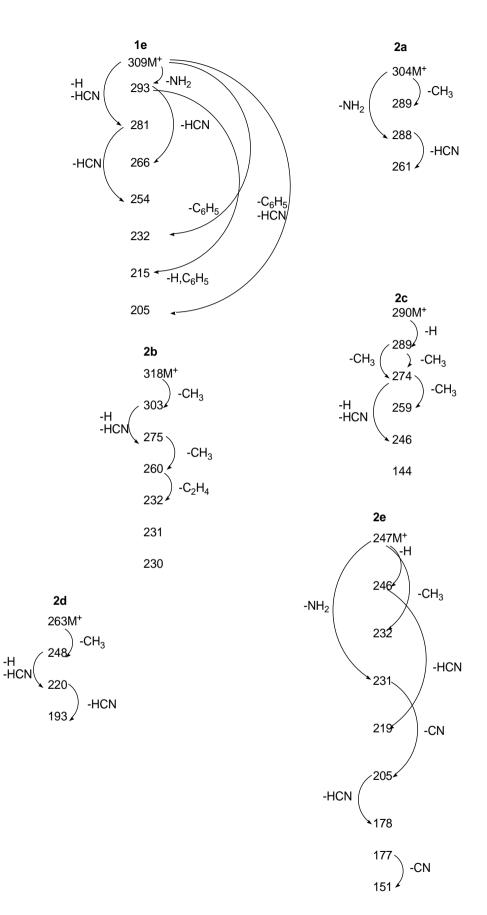
COMPOUND	R	R ¹
3a	C ₆ H ₅	Н
3b	C ₆ H ₅	CH₃
Зс	Furfuryl	CH ₃

Similar loss of CH_3 , HCHO and HCN was observed both methyl substituted compounds **1b** and **1d** from the molecular ion shown in **scheme -I.** A successive loss of phenyl groups were observed compound **1c** and fragmentation pattern of m-terphenyl compounds**1b** and **1c** is depicted in **scheme** –II.

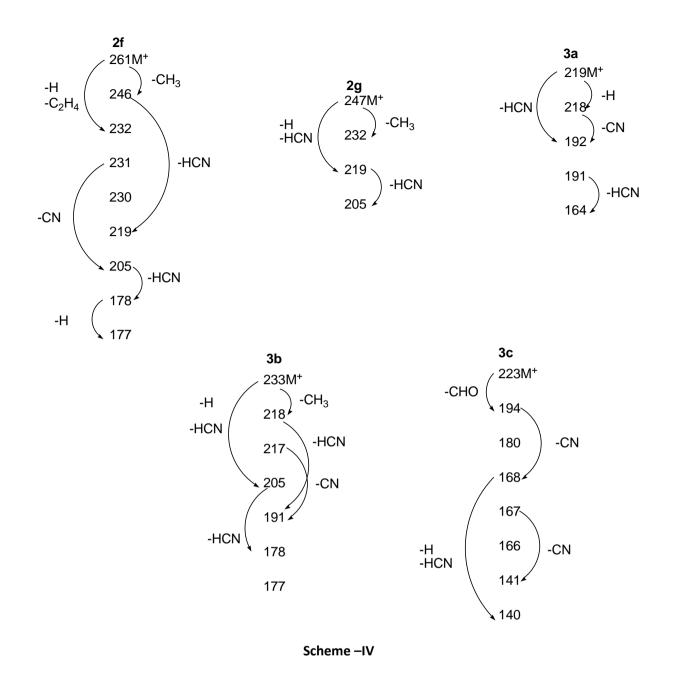




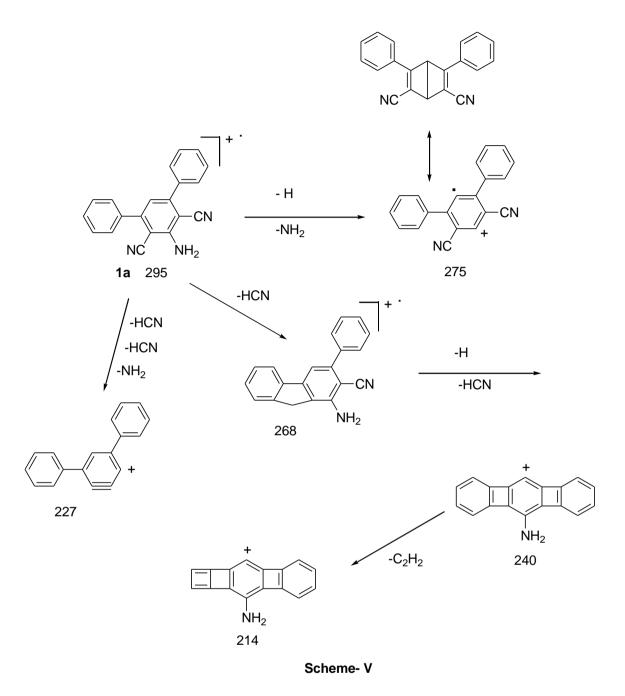
Scheme- II



Scheme –III



The representative fragmentation pattern of m- terphenyl compound **1a** is furnished in **scheme-V**.



Conclusion

Totally **15** m-terphenyl and biaryl derivatives were studied for their mass spectral fragmentation pattern. Molecular mass information and fragmentation pattern are easily obtained under the electron ionization (EI) condition. The most common loss of CN,NH₂,HCN,C₂H₄ groups were observed from all the compounds **1a-e**, **2a-g**, **3a-c**.Similar loss of CH₃, HCHO and HCN groups were identified both methyl substituted compounds **1b** and **1d** from the molecular ions. Successive lossof phenyl groups were observed in compound **1c** and the loss of CHO group is the characteristic of furfurylmoiety has been observed in compound **3c** (m/z 194).

Acknowledgment

We thank the DST, India for the financial support of this work and Prof V.T. Ramakrishnan for generous support for all my work.

References

1. Saednya. A, Hart.H, Synthesis, 1996, 1455-1459.

- 2. (a) Hart. H,Rajakumar, P. Tetrahedron,34,1995,4407-4413 (b) Rajakumar. P,Srisailas.M, Tetra hedron Lett,34,1993,5323 5326.
- 3. Rajakumar. P, Kannan. A, Tetrahedron let, 34, 1993, 4407-4409.
- 4. Bringmann.G, Walte,R, Weirich,R, Synthesis,1990,977-984.
- 5. Catellani. M, Motti. E, Minari. M, Chem. Comm, 2000, 157-160.
- a)Murugan.P, Shanmugasundaram.P, Ramakrishnan.VT, Venkatachalapathy. B,Srividya,N, Ramamurthy.P, Gunasekaran.K, Velmurugan.D,J.Chem.Soc.Perkin Trans.2,1998,999-1007;
 b)Shanmugasundaram.P, Murugan.P, Ramakrishnan. VT, Heteroatom Chemistry,7,.1996,17-23.
- 7. Murugan, P, Ramakishnan, VT, Srividya, N, Ramamurthy. P, Indian J. Chem, 37B, 1998, 737-742.
- 8. a)Murugan.P, Raghukumar. V,Ramakrishnan.VT,Synth.Commun,29,1999,3881-3886. b) Raghukumar. V, Murugan,P, Ramakrishnan.VT,Synth.Commun,31,2001,3497-3482.
