



International Journal of ChemTech Research CODEN (USA): IJCRGG ISSN : 0974-4290 Vol.6, No.4, pp 2427-2437, July-Aug 2014

Design, Synthesis of some Novel 1,3,4-Oxadiazole derivatives bearing Benzimidazole Nucleus and Biological evaluation of their possible *in-vitro* Anti-inflammatory and Antioxidant activity

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Abstract: Some novel 1,3,4-oxdiazole derivatives bearing benzimidazole nucleus were synthesized and their *invitro* anti-inflammatory and antioxidant activity by inhibition of protein denaturation screening, 2, 2-Diphenyl-1-Picryl Hydrazide screening methods respectively. The compound **7.XX** and **7.XXIX** were found to be highly active in low concentration and compounds **7.VIII**, **7.IX**, **7.X**, **7.XIX**, **7.XXII**, **7.XXIV** and **7.XXV** were found to be moderately active at higher concentration as compared to ascorbic acid in 2, 2-diphenyl-1-picryl hydrazide method. *In-vitro* anti-inflammatory by inhibition of protein denaturation method the compounds **7.VIII**, **7.XX**, **7.XXII** and **7.XXIX** were found to be highly active in low concentration and compounds **7.IX**, **7.X**, **7.XIX**, **7.XXIV**, **7.XXV** and **7.XXIII** were found to be moderately active at higher concentration and compounds **7.IX**, **7.X**, **7.XIX**, **7.XXIV**, **7.XXV** and **7.XXXIII** were found to be moderately active at higher concentration as compared to diclofenac sodium. **Key Words:** o-phenylenediamine, 4-cyanobenzaldehyde, imines intermediate, 1,3, 4-oxadiazole.

1. Introduction

The structural and therapeutic diversity coupled with commercial capability of small molecules has enthralled organic and medicinal chemists. There has been significant interest in the chemistry of oxadiazole ring systems, which is a core structure in various synthetic pharmaceuticals displaying a broad spectrum of biological activity. In the past few years, research for new non-steroidal anti-inflammatory agents has been reported that many compounds having a 1,3,4-oxadiazole derivatives are known for their anti-inflammatory activity.¹⁻³

Oxadiazole derivatives play a significant role in various pharmaceuticalapplications.⁴⁻⁶ As an important class of heterocyclic compound, 1,3,4-oxadiazoles show broad spectrum of biological activities.⁷⁻¹² Among these, a few differently substituted1,3,4-oxadiazoles have exhibited potent antitumor activities particularly.¹³⁻¹⁵ Of the various human diseases, cancer has proven to be one of the most intractable diseases to which humans are subjected, and as yet no practical and generally effective drugs or methods of control are available. Therefore, identification of novel potent, selective and less toxic anti-cancer agent's remains one of the most pressing health problems.¹⁶ Targeting tubulin in rapidly dividing tumor cells has been a well validated strategy for cancer therapy.^{17, 18} Benzimidazole derivatives are well known for their anti-inflammatory activity and more recently have been discovered to have anticancer effect.^{19, 20} Therefore, in the present research it was planned to incorporate the Benzimidazole moiety with 1,3,4-oxadiazole to have better antioxidant and anti-inflammatory activity.

2. Materials and methods

2.1. Chemistry

Melting points were determined by open capillary method and were uncorrected. The IR spectra (in KBr) were recorded on a Shimadzu IR Affinity-1 spectrophotometer. ¹H NMR and ¹³CNMR spectra were recorded on a Perkin–Elmer EM 300 MHz spectrometer using TMS as internal standard. The mass spectra were recorded on a JEOL JMS-D 300 spectrometer operating at 70 eV. Purity of the compounds was checked by TLC silica coated plates obtained from Merck.

General procedure for the preparation of 4-{1-[(4-acetyl-5-(substituted)-4, 5-dihydro-1, 3, 4-oxadiazol-2-yl) methyl]-5-nitro-1*H*-benzimidazol-2-yl} benzonitrile **6** (**I-IV**)

A mixture of imine intermediate **4** (**I-IV**) (0.01 mole) and excess of acetic anhydride (10 ml) was refluxed for 3-4 h. The acetic anhydride was distilled off and the residue was poured on to crushed ice. The solid thus obtained was collected by filtration, washed with water and recrystallized from ethanol. The purity of the product was confirmed by a single spot on TLC plate.

General procedure for the preparation of novel 4-{5-substituted-1-[(4,5-disubstituted)-4,5-dihydro-1,3,4-oxadiazol-2-yl}methyl]-5-nitro-1*H*-benzimidazol-2-yl}benzonitrile **7** (**V-XXXIII**).

To a solution of imine intermediate **5** (V-XXXIII) (0.01 mole) in ethanol (15 ml) and Chloramines-T (0.01 mole) was added. The reaction mixture was exposed to microwave at 300 W intermittently at 30 sec intervals for specified time. After complete conversion as indicated by TLC, the reaction mixture was cooled and digested with cold water.

3. Biological Evaluation

3.1. Antioxidant and free radical scavenging activity assays

3.1.1. DPPH assay

This experimental procedure was adapted from.²¹ In an ethanol solution of 2, 2-diphenyl-1-picrylhydrazyl (DPPH) radical, test compounds at different concentrations were added. The reaction mixtures were shaken vigorously and then kept in the dark for 30 min. The absorbance of the resulting solutions was measured in 1 cm cuvettes, using a UV/VIS spectrophotometer at 351nm against blank without DPPH. Decreasing of DPPH solution absorbance indicates an increase of DPPH radical scavenging activity. This activity is given as % DPPH radical scavenging that is calculated in the equation:

Demonstrate inhibition -	Absorbance of control - Absorbance of test	X 100
Percentage inhibition =	Absorbance of control	· · · · · · · · · · · · · · · · · · ·

The DPPH solution without sample solution was used as control. All tests were run in triplicate and averaged. Ascorbic acid was used as positive control in (Table 2 and Figure 2).

3.2. In vitro anti-inflammatory activity

3.2.1. Inhibition of protein denaturation method

Test solution (0.5 mL) consists of 0.45 mL of BSA (5% w/v aqueous solution) and 0.05 mL of different concentration of test solutions (50, 100, 150, 200 μ g/mL). Test control solution (0.5 mL) consists of 0.45 mL of BSA (5% w/v aqueous solution) and 0.05 mL of distilled water. Product control solution (0.5 mL) consists of 0.45 mL of distilled water and 0.05 mL of different concentration of test solutions (50, 100, 150, 200 μ g/mL). Standard solution (0.5 mL) consists of 0.45 mL of BSA (5% w/v aqueous solution) and 0.05 mL of different concentration of test solutions (50, 100, 150, 200 μ g/mL). Standard solution (0.5 mL) consists of 0.45 mL of BSA (5% w/v aqueous solution) and 0.05 mL different concentration of Diclofenac Sodium's (50, 100, 150, 200 μ g/mL). All the above solutions were adjusted to pH 6.3 using 1N hydrochloric acid. The samples were incubated at 37^oC for 20 min and the temperature was increased to keep the

samples at 57^oC for 3 min. After cooling, 2.5 mL of phosphate buffer saline was added to the above solutions. The absorbance was measured using UV Visible spectrophotometer at 417 nm.²² The percentage inhibition of protein denaturation was calculated as,

Percentage inhibition = O. D of Test Solution - O. D of Product Control O. D of Test Control X 100

The results were compared with Diclofenac Sodium in (Table 3 and Figure 3).

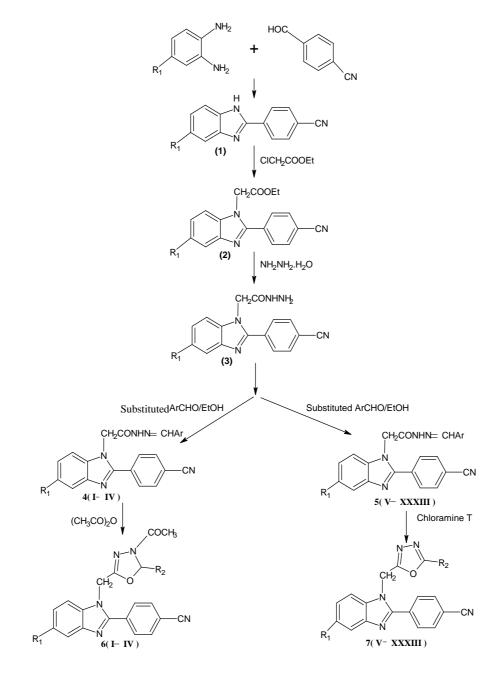


Figure 1: Scheme I Synthetic route for the preparation of the compounds.

Compound Code	R ₁	\mathbf{R}_2	R ₃	Molecular Formula	Molecular Weight	Yield (%)	Melting point (°C)	R _f value
6. I	-NO ₂	CI	-COCH ₃	C ₂₅ H ₁₇ ClN ₆ O ₄	500.89328	25	273-275	0.7311
6. II	-NO ₂		-COCH ₃	C ₂₅ H ₁₇ ClN ₆ O ₄	500.89328	30	270-272	0.7272
6. III	-NO ₂	HO	-COCH ₃	$C_{25}H_{18}N_6O_6$	498.44702	35	213-215	0.7647
6. IV	-NO ₂	-	-COCH ₃	$C_{25}H_{17}BrN_6O_4$	545.34428	32	210-212	0.6860
7. V	-NO ₂		-	$C_{23}H_{13}ClN_6O_3$	456.84072	30	269-271	0.7701
7. VI	-NO ₂		-	C ₂₃ H ₁₃ ClN ₆ O ₃	456.84072	26	274-276	0.7294
7. VII	-NO ₂	H ₂ N	-	$C_{23}H_{15}N_7O_3$	437.4103	20	243-245	0.7311
7. VIII	-NO ₂	CI	-	C ₂₃ H ₁₃ ClN ₆ O ₃	456.84072	28	275-277	0.7272
7. IX	-Cl	Br	-	C ₂₃ H ₁₃ BrClN ₅ O	490.73922	31	209-211	0.7647
7. X	-NO ₂	-CI	-	$C_{23}H_{12}Cl_2N_6O_3$	491.28578	30	276-278	0.6860
7. XI	-NO ₂	Ностори	-	$C_{23}H_{14}N_6O_5$	454.39446	33	211-213	0.7701
7. XII	-NO ₂	Br	-	$\mathrm{C}_{23}\mathrm{H}_{13}\mathrm{BrN}_{6}\mathrm{O}_{3}$	501.29172	28	211-213	0.7294

 Table 1: Characterization Data Scheme I

7. XIII	-NO ₂	—————Br	-	$C_{23}H_{13}BrN_6O_3$	501.29172	20	183-185	0.7311
7. XIV	-NO ₂	-CH ₂ -Br	-	$C_{24}H_{15}BrN_6O_3$	515.3183	26	208-210	0.7272
7. XV	$-NO_2$	H ₃ C	-	$C_{24}H_{16}N_6O_3$	436.42224	30	193-195	0.7647
7. XVI	-NO ₂	H ₃ CO	-	$C_{24}H_{16}N_6O_4$	452.42164	35	223-225	0.6860
7. XVII	-NO ₂		-	C ₂₃ H ₁₂ BrN ₇ O ₅	546.28928	23	256-258	0.7701
7. XVIII	-NO ₂	Br OH	-	C ₂₃ H ₁₃ BrN ₆ O ₄	517.29112	31	243-245	0.7294
7. XIX	-NO ₂	HO F F	-	$C_{23}H_{12}F_2N_6O_4$	474.3759864	20	254-256	0.7311
7. XX	-NO ₂		-	$C_{23}H_{12}Cl_2N_6O_3$	491.28578	36	285-287	0.7272
7. XXI	-NO ₂		-	C ₂₃ H ₁₂ ClN ₇ O ₅	501.83828	33	276-278	0.7647
7. XXII	-NO ₂	Br ————————————————————————————————————	-	$C_{24}H_{12}BrF_3N_6O_3$	569.2896896	25	252-254	0.6860
7. XXIII	$-NO_2$		-	C ₂₃ H ₁₂ ClN ₇ O ₅	501.83828	31	263-265	0.7701
7. XXIV	-NO ₂		_	C ₂₅ H ₁₇ ClN ₆ O ₅	516.89268	30	224-226	0.7294
7. XXV	-NO ₂	H ₃ CH ₂ CO ————————————————————————————————————	-	$C_{26}H_{20}N_6O_5$	496.4742	19	214-216	0.7311

7. XXVI	-NO ₂		-	$C_{27}H_{20}N_6O_6$	524.4843	20	219-221	0.7272
7. XXVII	-NO ₂	OCH ₂ CH ₂ CH ₂ Br	-	$C_{26}H_{19}BrN_6O_4$	559.37086	25	223-225	0.7647
7. XXVIII	-NO ₂		-	$C_{28}H_{20}N_6O_5$	520.4956	23	222-224	0.6860
7. XXIX	-NO ₂		-	$C_{25}H_{18}N_6O_6$	498.44702	19	212-214	0.7701
7. XXX	-NO ₂		-	$C_{23}H_{12}N_8O_8$	528.39018	34	262-264	0.7294
7. XXXI	-NO ₂	HO	-	$C_{31}H_{30}N_6O_4$	550.6077	19	221-223	0.7311
7. XXXII	-NO ₂	H ₂ CHCH ₂ C OH	-	$C_{26}H_{18}N_6O_4$	478.45892	36	226-228	0.7272
7. XXXIII	-NO ₂		-	$C_{26}H_{16}N_8O_3$	488.45704	37	222-224	0.7647

Compound Code	IC ₅₀ Value(µg/ml)	IC ₅₀ Value±SEM (µg/ml)
7.VIII	160	160±0.9614
7.IX	300	300±0.9865
7.X	220	220±1.0425
7. XIX	165	165±0.67
7. XX	116	116±1.025
7. XXII	147	147±0.96
7. XXIV	148	148 ± 1.052
7. XXV	137	137±1.16
7. XXIX	103	103±0.96
Ascorbic acid	109	109±0.7296

Table 2: The IC₅₀±SEM (standard error of mean) Values of Ascorbic Acid and Compounds (6.I to 7.XXXIII)

Values represent the mean of triplicates.

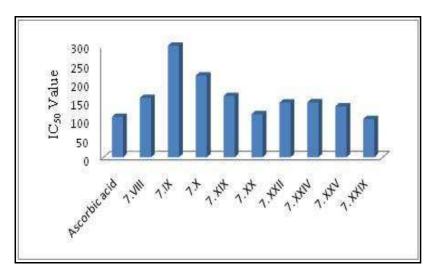


Figure 2: The IC₅₀ values of compounds (6.I to 7.XXXIII) and ascorbic acid (standard) in 2, 2-Diphenyl-1-Picryl Hydrazide screening of antioxidant activity. The results are (mean \pm SEM) of three experiments, performed in triplicate.

Compound Code	IC ₅₀ Value(µg/ml)	IC50 Value±SEM(µg/ml)
7. V	137.4	137.4±0.881917
7. VI	119.8	119.8±1.527525
7.VIII	64.98	64.98±1.154701
7.X	76.77	76.77±0.5773503
7. XIX	86.48	86.48±0.5773503
7. XX	67.59	67.59±0.8819171
7. XXII	61.02	61.02±1.20185
7. XXIV	99.56	99.56±1.154701
7. XXV	99.16	99.16±0.8819171
7. XXVIII	100	100±0.5773503
7. XXIX	67.5	67.5±0.9865
Diclofenac Sodium	57.08	57.08±0.8819

Values represent the mean of triplicates.

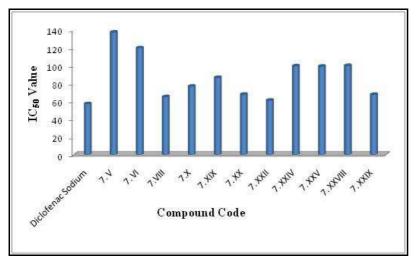


Figure 3: The IC₅₀ values of compounds (6.I to 7.XXXIII) and diclofenac sodium (standard) in inhibition of protein denaturation screening of *In-Vitro* Anti-Inflammatory activity. The results are (mean \pm SEM) of three experiments, performed in triplicate.

4. Results and discussion

For the synthesis of the target compounds the reaction sequences outlined in Scheme 1, were followed. 4-{5-substituted-1-[(4, 5-disubstituted)-4, 5-dihydro-1, 3, 4-oxadiazol-2-yl) methyl]-5-nitro-1*H*-benzimidazol-2-yl} benzonitrile **7** (V-XXXIII) had been synthesized employing conventional techniques. Substituted o-phenylene diamine was reacted with appropriately substituted 4-cyanobenzaldehydes in the presence of sodium metabisulfite to furnish substituted 2-(4-Cyanophenyl)-1H-benzimidazoles (1). These substituted 2-(4-Cyano phenyl)-1Hbenzimidazoles were further treated with ethyl chloroacetate in KOH/DMSO gave the N-alkylated product, (2-(4cyanophenyl)-benzimidazol-1-yl)-acetic acid ethyl esters (2). To endow2-(4-cyanophenyl)-benz imidazol-1-yl)acetic acid hydrazides (3) reaction were occurred between Hydrazine hydrate and the esters (2). A mixture of 2-(4cyanophenyl)-benzimidazol-1-yl)-acetic acid hydrazides and respective aldehydes were react to generate imines intermediates **4** (**I-IV**) **& 5** (**V-XXXIII**). The imines intermediate and excess of acetic anhydride react to produce $4-\{1-[(4-acetyl-5-(substituted)-4, 5-dihydro-1,3,4-oxadiazol-2-yl) methyl]-5-nitro-1$ *H* $-benz imidazol-2-yl} benzo$ nitrile**6**(**I-IV**) in Figure 1. The imines intermediate in ethanol and Chloramines-T react to produce the products**7** (**V-XXXIII**) in Figure 1.

7. V: 4-{1-[(5-(2-chlorophenyl)-1,3,4-oxadiazol-2-yl) methyl]-5-nitro-1H-benzimidazol-2-yl} benzonitrile

Yield 30%; m.p.:269-271; IR (cm⁻¹): 3062 (Ar-C-H), 2210 (C≡N), 1640 (C=N), 1596, 1473, 1439 (C=C), 1352 (N=O), 1331 (C-N), 1242 (C-O-C), 733 (C-Cl)

¹H NMR data (CDCl₃) δ (ppm) and ¹³C NMR data (ppm): 5.126 (s, 2H, -CH₂), 7.361-8.437 (m, 11H, aromatic protons), ¹³C:49, 112, 114, 119, 120, 126, 129, 130, 131, 132, 134, 136, 141, 144, 145, 156, 166, 167; MS: m/z 457 (M⁺)

7. VI: 4-{1-[(5-(3-chlorophenyl)-1,3,4-oxadiazol-2-yl) methyl]-5-nitro-1H-benzimidazol-2-yl} benzonitrile

Yield 26%; m.p.:274-276; IR (cm⁻¹): 3062 (Ar-C-H), 2210 (C≡N), 1640 (C=N), 1596, 1473, 1439 (C=C), 1352 (N=O), 1331 (C-N), 1242 (C-O-C), 733 (C-Cl)

¹H NMR data (CDCl₃) δ (ppm) and ¹³C NMR data (ppm): 5.126 (s, 2H, -CH₂), 7.332-8.437 (m, 11H, aromatic protons), ¹³C:49, 112, 114, 119, 120, 127, 128, 129, 131, 133, 134, 136, 141, 144, 145, 158, 165, 167; MS: m/z 457 (M⁺)

7. VIII: 4-{1-[(5-(4-chlorophenyl)-1,3,4-oxadiazol-2-yl) methyl]-5-nitro-1*H*-benzimidazol-2-yl} benzonitrile. Yield 28%; m.p.:275-277; IR (cm⁻¹): 3062 (Ar-C-H), 2210 (C \equiv N), 1640 (C=N), 1596, 1473, 1439 (C=C), 1352 (N=O), 1331 (C-N), 1242 (C-O-C), 733 (C-Cl). ¹H NMR data (CDCl₃) δ (ppm) and ¹³C NMR data (ppm): 5.126 (s, 2H, -CH₂), 7.362-8.43 (m, 11H, aromatic protons), ¹³C:49, 112, 114, 119, 120, 127, 129, 130, 131, 134, 136, 137, 141, 145, 156, 164, 167; MS: m/z 457 (M⁺)

7. IX: 4-{1-[(5-(3-bromophenyl)-1,3,4-oxadiazol-2-yl) methyl]-5-chloro-1*H*-benzimidazol-2-yl} benzonitrile. Yield 31%; m.p.:209-211; IR (cm⁻¹): 3062 (Ar-C-H), 2210 (C \equiv N), 1640 (C=N), 1596, 1473, 1439 (C=C), 1331 (C-N), 1242 (C-O-C), 733 (C-Cl), 575 (C-Br). ¹H NMR data (CDCl₃) δ (ppm) and ¹³C NMR data (ppm): 5.126 (s, 2H, -CH₂), 7.137-8.11 (m, 11H, aromatic protons), ¹³C: 49, 112, 114, 119, 123, 124, 125, 128, 130, 131, 132, 134, 137, 144, 158, 164, 167; MS: m/z 491, 493 (M⁺)

7. X: 4-{1-[(5-(3, 4-dichlorophenyl)-1,3,4-oxadiazol-2-yl) methyl]-5-nitro-1*H*-benzimidazol-2-yl} benzonitrile. Yield 30%; m.p.:276-278; IR (cm⁻¹): 3062 (Ar-C-H), 2210 (C \equiv N), 1640 (C=N), 1596, 1473, 1439 (C=C), 1352 (N=O), 1331 (C-N), 1242 (C-O-C), 733 (C-Cl). ¹H NMR data (CDCl₃) δ (ppm) and ¹³C NMR data (ppm): 5.126 (s, 2H, -CH₂), 7.36-8.43 (m, 10H, aromatic protons), ¹³C: 49, 112, 114, 119, 120, 125, 128, 131, 132, 133, 134, 136, 141, 144, 145, 158, 165, 167; MS: m/z 491 (M⁺)

7. XIX:4-{1-[(5-(4,5-difluoro-2-(1,3,4-oxadiazol-2-yl)phenol)methyl]-5-nitro-1*H*-benzimidazol-2-yl} benzonitrile. Yield 20%; m.p.:254-256; IR (cm⁻¹): 3655 (O-H), 3062 (Ar-C-H), 2210 (C=N), 1640 (C=N), 1596, 1473, 1439 (C=C), 1356 (C-F), 1352 (N=O), 1331 (C-N), 1242 (C-O-C). ¹H NMR data (CDCl₃) δ (ppm) and ¹³C NMR data (ppm): 5.126 (s, 2H, -CH₂), 6.84-8.43 (m, 9H, aromatic protons), 4.71 (s, 1H, OH), ¹³C: 49, 101, 110, 112, 114, 116, 119, 120, 131, 134, 136, 141, 144, 145, 151, 155, 158, 167, 168; MS: m/z 474 (M⁺)

7. XX: 4-{1-[(5-(3, 5-dichlorophenyl)-1,3,4-oxadiazol-2-yl) methyl]-5-nitro-1*H*-benzimidazol-2-yl} benzonitrile. Yield 36%; m.p.:285-287; IR (cm⁻¹): 3062 (Ar-C-H), 2210 (C \equiv N), 1640 (C=N), 1596, 1473, 1439 (C=C), 1352 (N=O), 1331 (C-N), 1242 (C-O-C), 733 (C-Cl). ¹H NMR data (CDCl₃) δ (ppm) and ¹³C NMR data (ppm): 5.126 (s, 2H, -CH₂), 7.36-8.43 (m, 9H, aromatic protons), ¹³C: 49, 112, 114, 119, 120, 127, 129, 131, 134, 135, 136, 139, 141, 144, 145, 158, 164, 167; MS: m/z 491 (M⁺)

7. XXII: 4-{1-[(5-(2-bromo-4-(trifluoromethyl) phenyl)-1,3,4-oxadiazol-2-yl) methyl]-5-nitro-1*H*-benzimidazol-2-yl} benzonitrile. Yield 25%; m.p.:252-254; IR (cm⁻¹): 3062 (Ar-C-H), 2210 (C=N), 1640 (C=N), 1596, 1473, 1439 (C=C), 1356 (C-F), 1352 (N=O), 1331 (C-N), 1242 (C-O-C), 575 (C-Br). ¹H NMR data (CDCl₃) δ (ppm) and ¹³C NMR data (ppm): 5.126 (s, 2H, -CH₂), 7.36-8.43 (m, 10H, aromatic protons), ¹³C: 49, 112, 114, 119, 120, 124, 126, 127, 128, 131, 134, 135, 136, 141, 145, 158, 164, 167; MS: m/z 569, 571 (M⁺)

7. XXIV: 4-{1-[(5-(2-chloro-4, 5-dimethoxyphenyl)-1, 3, 4-oxadiazol-2-yl) methyl]-5-nitro-1*H*-benzimidazol-2-yl} benzonitrile. Yield 30%; m.p.:224-226; IR (cm⁻¹): 3062 (Ar-C-H), 2210 (C=N), 1640 (C=N), 1596, 1473, 1439 (C=C), 1352 (N=O), 1331 (C-N), 1242 (C-O-C), 733 (C-Cl). ¹H NMR data (CDCl₃) δ (ppm) and ¹³C NMR data (ppm): 5.126 (s, 2H, -CH₂), 3.77 (s, 6H, -CH₃), 7.36-8.43 (m, 9H, aromatic protons), ¹³C: 49, 56, 112, 113, 114, 117, 118, 119, 120, 127, 131, 134, 136, 144, 145, 151, 158, 165, 167; MS: m/z 517 (M⁺)

7.XXV:4-{1-[(5-(2-ethoxy-4-methoxyphenyl)-1,3,4-oxadiazol-2-yl)methyl]-5-nitro-1*H*-benzimidazol-2-yl} benzonitrile. Yield 19%; m.p.:214-216; IR (cm⁻¹): 3062 (Ar-C-H), 2849 (C-H), 2210 (C=N), 1640 (C=N), 1596, 1473, 1439 (C=C), 1352 (N=O), 1331 (C-N), 1305 (C-C), 1242 (C-O-C). ¹H NMR data (CDCl₃) δ (ppm) and ¹³C NMR data (ppm): 4.148-5.126 (s, 4H, -CH₂), 1.45-3.82 (s, 6H, -CH₃), 6.62-8.43 (m, 10H, aromatic protons), ¹³C: 15, 49, 56, 64, 97, 105, 108, 112, 114, 119, 120, 127, 131, 134, 136, 141, 144, 145, 157, 158, 164, 166, 167; MS: m/z 496 (M⁺)

7. XXVIII: 4-{1-[(5-[3-ethoxy-4-(prop-2-yn-1-yloxy) phenyl]-1, 3, 4-oxadiazol-2-yl) methyl]-5-nitro-1*H*-benzimidazol-2-yl} benzonitrile. Yield 23%; m.p.:222-224; IR (cm⁻¹): 3062 (Ar-C-H), 2849 (C-H), 2119 (C=C), 2210 (C=N), 1640 (C=N), 1596, 1473, 1439 (C=C), 1352 (N=O), 1331 (C-N), 1305 (C-C), 1242 (C-O-C). ¹H NMR data (CDCl₃) δ (ppm) and ¹³C NMR data (ppm): 2.41 (s, 1H, -CH), 4.09-5.126 (s, 6H, -CH₂), 1.488 (s, 3H, -CH₃), 6.87-8.43 (m, 10H, aromatic protons), ¹³C: 15, 49, 57, 65, 76, 79, 110, 111, 112, 114, 119, 120, 123, 131, 134, 136, 141, 144, 145, 149, 157, 158, 163, 167; MS: m/z 520 (M⁺)

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7.XXIX:4-{1-[(5-[2,3-dimethoxy-6-(1,3,4-oxadiazol-2-yl)phenol]methyl]-5-nitro-1*H*-benzimidazol-2-yl} benzonitrile. Yield 19%; m.p.:212-214; IR (cm⁻¹): 3655 (O-H), 3062 (Ar-C-H), 2849 (C-H), 2210 (C \equiv N), 1640 (C=N), 1596, 1473, 1439 (C=C), 1352 (N=O), 1331 (C-N), 1242 (C-O-C). ¹H NMR data (CDCl₃) δ (ppm) and ¹³C NMR data (ppm): 5.126 (s, 2H, -CH₂), 3.75-3.85 (s, 6H, -CH₃), 6.70-8.47 (m, 9H, aromatic protons), 5.44 (s, 1H, OH), ¹³C: 49, 57, 60, 95, 107, 112, 114, 119, 120, 121, 131, 134, 136, 141, 142, 144, 145, 149, 157, 158, 167; MS: m/z 498 (M⁺)

The compound 7.XX (116±1.025) μ g/ml and 7.XXIX (103±0.96) μ g/ml were found to be highly active in low concentration and compounds 7.VIII (160±0.9614) μ g/ml, 7.IX (300±0.9865) μ g/ml, 7.X (220±1.0425) μ g/ml, 7.XIX (165±0.67) μ g/ml, 7.XXII (147±0.96) μ g/ml, 7. XXIV (148±1.052) μ g/ml and 7.XXV (137±1.16) μ g/ml were found to be moderately active at higher concentration as compared to ascorbic acid (109±0.7296) μ g/ml in 2, 2-diphenyl-1-picryl hydrazide method. *In-vitro* anti-inflammatory by inhibition of protein denaturation method the compounds 7.VIII (64.98±1.154701) μ g/ml, 7.XX (67.59±0.8819171) μ g/ml, 7.XXII (61.02±1.20185) μ g/ml and 7.XXIX (67.5±0.9865) μ g/ml were found to be highly active in low concentration and compounds 7.V (137.4±0.881917), 7.VI (119.8±1.527525), 7.X (76.77±0.5773503) μ g/ml, 7.XIX (86.48±0.5773503) μ g/ml, 7. XXIV (99.56±1.154701) μ g/ml, 7.XXV (99.16±0.8819171) μ g/ml and 7.XXVIII (100±0.5773503) μ g/ml were found to be moderately active at higher concentration as compared to diclofenac sodium (57.08±0.8819) μ g/ml.

5. Conflict of interest statement

We declare that we have no conflict of interest

6. Acknowledgements

The authors are thankful to the authority of Sumandeep Vidyapeeth University and Department of Pharmacy for providing necessary facilities.

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