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NOx Formation and Control: On Thermochemistry of NNH

TonyeK. Jack*

Seltrolene Consulting, Port Harcourt, 50004, Rivers State, Nigeria.

Abstract: The NNH species mechanism pathway is an important consideration in currentNOx combustion pollutants formationat high temperatures, and low residence time controlmodelling. However, no detailed Table of NNH thermochemical properties exists in the open chemicals databases. Attempt is made in this paper at developing a detailed NNH thermochemical data Table. The 7-term NASA polynomial coefficients for NNH are applied. No 9-term coefficients yet cited. The thermochemical quantities tabulations follow the format of Barin as reported in thesurvey by Jacobsonof available source compendia of thermochemical data. TwoMethods of solutions by additivity for the energies of formations, by, linking dynamically, related and consistent listings in terms of the choice of stable molecule reference elements, lead to similar results.NASA coefficients were used to precision, but round-off of final listed results is to 3-decimals.Curve-fitted equations of equilibrium constant for the two methods, with predictions accurate to within maximum deviations of ± 0.50 percent, and ±0.02-percent from tabulated data, and averaged absolute uncertainties of 0.08percent, and 0.01-percent in the 1000 -to-6000 K temperature range, respectively are presented. **Keywords:**Thermochemistry;NNH;PollutantFormation;NOx Chemistry; CombustionReactions.

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