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# Activated carbon-doped with iron oxide nanoparticles ( $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> NPs) preparation: particle size, shape, and impurity

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**Abstract :** Inspired by the development of an adsorbent material with a high capacity for chlorinated gas treatment, an  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> nanoparticle deposited activated carbon adsorbent was successfully synthesized. The  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> NPs synthesis was done by a facile chemical precipitation method using sodium hydroxide (NaOH) as a precipitant agent. The variation of the molar ratio of the reactant and precipitant (i.e. 1:1, 1:1.5, 1:2 by mole) and of the precipitating temperature (i.e. 50, 70, 90°C) were explored. The physical and chemical characteristics of the synthesized samples were examined using various techniques; Transmission Electron Microscope (TEM), Brunauer-Emmett-Teller analysis (BET), Thermogravimetry analysis (TGA), Fourier Transform Infra-Red (FT-IR) and Ultraviolet-Visible spectrophotometer. The result shows that two synthesized conditions - 1:1 by mole (FeCl<sub>3</sub>:NaOH) at 70°C and 1:1.5 by mole (FeCl<sub>3</sub>:NaOH) at 90°C - were allowed to produce the smallest size of  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> NPs approximately 10 nm. Consequently, it gave the highest specific surface area of ~110 m<sup>2</sup>/g. With a higher FeCl<sub>3</sub>: NaOH molar ratio and a higher precipitating temperature, the synthesized Fe<sub>2</sub>O<sub>3</sub> NPs formed a more oval shape with a finer surface. Due to the insufficient purification, unfortunately, an impurity caused by sodium salt was detected in the amount of 5-10 wt.%. The minimum amount of 75 wt.%  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> NPs coated on the activated carbon was observed. The production yield of the synthesized  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> and  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>/GAC samples was also reported.

**Keywords :** Iron oxide ( $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>), nanoparticle, activated carbon, synthesis, characterization.

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