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Synthesis of Hydroxy-sodalite from Fine Fractions of Sandy Clay Loam Soil (Natural Aluminosilicate)

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Abstract : Hydroxy-sodalite zeolites of different morphologies were synthesized using fine fractions of sandy clay loam soil (Natural aluminosilicate) by alkaline fusion method prior to conventional alkaline hydrothermal reaction (90°C for 24 hours). Mineralogical, textural properties and elemental analysis of the products were probed by employing x-ray diffraction (XRD),Fourier transform infrared spectroscopy (FTIR), scanning electron microscopy coupled with energy dispersive spectroscopy (SEM-EDS), BET surface area and the cationexchange capacity (CEC).XRD and FTIR analysis confirmed the formation of Hydroxy-sodalite particles under all the experimental conditions. The BET surface area and total pore volume of the Hydroxy-sodalite formed varied from $(31 - 51) \text{ m}^2 / \text{g}$ and $(0.17 - 0.21) \text{ cm}^3 / \text{g}$ respectively. Cation exchange capacity as high as 203.06 meq / 100 g (low Si/Al ratio ≈ 1.02) was obtained at an optimum condition of NaOH / raw material ratio of 1.2 and a fusion temperature of 600°C.

Keywords : Sandy clay loam soil, Hydroxy-sodalite, synthesis, alkaline fusion.

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